

General Chemistry 1(CHEM101) Mid-Term Exam 1  
Duration: 1h 30 min

Fall2021

Student Name: _____	Student ID: _____
Professor's Name: _____	Section: L_____

- The exam contains 30 Multiple Choice Questions (Each question is 2 points, 60 points total)
- Answers should be entered in the "Answers Sheet" (page 4)
- A Simple Periodic Table is provided with the exam copy
- Avogadro number:  $N_A = 6.022 \times 10^{23}$  atoms/mol

A1. Select the correct increasing order (from small to large) of the following prefixes:

Gm, mm, pm, Mm, km (G: giga, m: milli, p: pico, M: mega, k: kilo)

- A) pm, mm, km, Mm, Gm      B) mm, km, pm, Mm, Gm      C) km, pm, Mm, mm, Gm  
D) pm, Mm, km, mm, Gm      E) pm, mm, km, Gm, Mm

$$\begin{aligned} 1\text{Gm} &= 10^9 \text{ m} \\ 1 \text{ mm} &= 10^{-3} \text{ m} \\ 1 \text{ pm} &= 10^{-12} \text{ m} \\ 1 \text{ Mm} &= 10^6 \text{ m} \\ 1 \text{ km} &= 10^3 \text{ m} \\ \Rightarrow \text{pm} &< \text{mm} < \text{km} < \text{Mm} < \text{Gm} \end{aligned}$$

A2. What is the temperature of  $-80^\circ\text{C}$  in Kelvin (K)?

- A) + 80K      B) 353K      C) 180K  
D) -353K      E) 193K

$$T(K) = T(^{\circ}\text{C}) + 273 = -80 + 273 = 193 \text{ K}$$

A3. The density of aluminum is  $2.70 \text{ g/cm}^3$ . A mass of 3.32 g aluminum has a volume of:

- A)  $6.02 \text{ cm}^3$       B)  $0.813 \text{ cm}^3$       C)  $8.96 \text{ cm}^3$   
D)  $1.23 \text{ cm}^3$       E)  $3.32 \text{ cm}^3$

$$V = \frac{m(g)}{d\left(\frac{\text{g}}{\text{cm}^3}\right)} = \frac{3.32 \text{ g}}{2.70 \frac{\text{g}}{\text{cm}^3}} = 1.23 \text{ cm}^3$$

A4. How many significant figures in the following measurement 0.031070:

- A) 3      B) 4      C) 5  
D) 6      E) 7

For a number less than one (< 1) all the zeros to the right of the first non-zero are significant: 0.031070 has 5 significant figures.

A5. Give the correct number of significant figures to the problem below:  $4.56 - 4.5512$

- A) 1 significant figure      B) 2 significant figures      C) 3 significant figures  
D) 4 significant figures      E) 5 significant figures

# Example Of Chem101 Exam Solutions

**Rachel S Tattersall**



**Example Of Chem101 Exam Solutions:**

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