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In the SEARCH bar type: stoichiometry ... select the 1st gizmo titled STOICHIOMETRY that appears.The first Gizmo shows this equation: $\text{Fe}_2\text{O}_3 + 3\text{CO} \rightarrow 2\text{Fe} + 3\text{CO}_2$

There is a Question to solve under the equation.

Here are the features that will help you solve each problem:

- You have a set of 16 tiles containing various conversion factors. Any of these tiles can be dragged down and placed on the green bar to solve the problem. The starting tile is already in place. It cannot be moved.
- When you place a tile correctly, you will see a red line that cancels units for you.
- The tiles can be flipped if the units do not cancel properly. Click the box, "Flip Tile" after you have dragged a tile down if you want to flip it.
- You can always check how the units are going. Click the box, "Show Units".
- When you have the right tiles in place, a box appears – "Show Numerical Result". Click the box after you have calculated the answer to see if you are correct.
- The "Reset" button moves all tiles away if you want to start over again. The "New Problem" button gives you a different problem with the same equation. "New Equation" button changes the balanced equation.

Explore these features for several minutes until you understand how to control them. Then do a minimum of 10 problems. (Each reaction has 4 or 5 problems, and there are at least 7 different equations.) Space to record your work is attached.

Grading rubric:	10 problems done =	20 points
	9 problems done =	18 points
	8 problems done =	16 points
	6-7 problems done =	14 points
	5 or fewer =	no credit until at least 6 are done

The last 5 problems are worth 1 point each to total 25 points for this assignment

Here is what to do:

Copy the Balanced Equation you are using: $\text{Fe}_2\text{O}_3 + 3\text{CO} \rightarrow 2\text{Fe} + 3\text{CO}_2$ What information is Given? (This information is on the first tile.) 1.75 moles Fe_2O_3 What information is Wanted? (What unit & name are you looking for?) moles CO_2

Copy the complete set of tiles that you used to solve this problem. Use a calculator to solve the problem, and check it in the Gizmo.

I have done a problem from the first reaction for you

1.75 mol Fe_2O_3	$\frac{3 \text{ mol CO}}{1 \text{ mol Fe}_2\text{O}_3}$	=	5.25 mol CO
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Do 10 problems on your own, or in a small group. You may access the Gizmo from home by using the URL and logon information provided. Keep practicing – each Gizmo problem will let you know when you have the correct set of factors.

Explorelearning Gizmo Answer Key Stoichiometry

Rachel S Tattersall



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Table of Contents Explorelearning Gizmo Answer Key Stoichiometry

1. Understanding the eBook Explorelearning Gizmo Answer Key Stoichiometry
 - The Rise of Digital Reading Explorelearning Gizmo Answer Key Stoichiometry
 - Advantages of eBooks Over Traditional Books
2. Identifying Explorelearning Gizmo Answer Key Stoichiometry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Explorelearning Gizmo Answer Key Stoichiometry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Explorelearning Gizmo Answer Key Stoichiometry
 - Personalized Recommendations
 - Explorelearning Gizmo Answer Key Stoichiometry User Reviews and Ratings
 - Explorelearning Gizmo Answer Key Stoichiometry and Bestseller Lists
5. Accessing Explorelearning Gizmo Answer Key Stoichiometry Free and Paid eBooks
 - Explorelearning Gizmo Answer Key Stoichiometry Public Domain eBooks
 - Explorelearning Gizmo Answer Key Stoichiometry eBook Subscription Services
 - Explorelearning Gizmo Answer Key Stoichiometry Budget-Friendly Options

6. Navigating Explorelearning Gizmo Answer Key Stoichiometry eBook Formats
 - ePub, PDF, MOBI, and More
 - Explorelearning Gizmo Answer Key Stoichiometry Compatibility with Devices
 - Explorelearning Gizmo Answer Key Stoichiometry Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Explorelearning Gizmo Answer Key Stoichiometry
 - Highlighting and Note-Taking Explorelearning Gizmo Answer Key Stoichiometry
 - Interactive Elements Explorelearning Gizmo Answer Key Stoichiometry
8. Staying Engaged with Explorelearning Gizmo Answer Key Stoichiometry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Explorelearning Gizmo Answer Key Stoichiometry
9. Balancing eBooks and Physical Books Explorelearning Gizmo Answer Key Stoichiometry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Explorelearning Gizmo Answer Key Stoichiometry
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Explorelearning Gizmo Answer Key Stoichiometry
 - Setting Reading Goals Explorelearning Gizmo Answer Key Stoichiometry
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Explorelearning Gizmo Answer Key Stoichiometry
 - Fact-Checking eBook Content of Explorelearning Gizmo Answer Key Stoichiometry
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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