

Observations

	1	Distilled H ₂ O	--	non
HC ₂ H ₃ O ₂ (aq): acetic acid	2	Tap water	--	weak
	3	NaCl	5	strong
	4	CuSO ₄	5	strong
HCl(aq): hydrochloric acid	5	Sugar	5	non
	6	Alcohol	5	non
NaOH(aq): sodium hydroxide	7	HCl	5	strong
	8	HCl	1	strong
CuSO ₄ (aq): copper(II) sulfate	9	HC ₂ H ₃ O ₂	5	weak
	10	HC ₂ H ₃ O ₂	1	weak
	11	NaOH	5	strong
	12	NaOH	1	strong
	13	HC ₂ H ₃ O ₂	17.4	non

Answers

1. An electrolyte is a compound that, in aqueous solution, conducts electricity. As liquids: salts, bases. When aq: salts, acids, bases.
2. HCl(aq): H⁺(aq) + Cl⁻(aq)
NaOH(aq): Na⁺(aq) + OH⁻(aq)
CuSO₄(aq): Cu²⁺(aq) + SO₄²⁻(aq)
3. Acid: a compound that ionizes in water to form hydrogen ions. Base: an ionic hydroxide that dissociates in water to produce hydroxide ions

Lab Conductivity Quizz With Answers

VM Jensen



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