

Isotopes Practice Set

- What does "atomic number" mean?
- What does the number next to isotopes mean?
- How do you tell isotopes apart?

For each of the following isotopes, write the number of protons, neutrons, and electrons.

	Chlorine-35	Chlorine-37
# of protons	17	17
# of neutrons	18	20
# of electrons	17	17

	Magnesium-24	Magnesium-26
# of protons	12	12
# of neutrons	12	14
# of electrons	12	12

	Carbon-12	Carbon-13
# of protons	6	6
# of neutrons	6	7
# of electrons	6	6

	Sulfur-32	Sulfur-34
# of protons	16	16
# of neutrons	16	18
# of electrons	16	16

Fill in the isotopes names including atomic numbers, including atomic numbers from the chart. Use your periodic table and the information provided.

	Isotope: $^{12}_6\text{C}$	Isotope: $^{16}_8\text{O}$
# of protons	6	8
# of neutrons	6	8
# of electrons	6	8

	Iron: $^{56}_{26}\text{Fe}$	Iron: $^{58}_{26}\text{Fe}$
# of protons	26	26
# of neutrons	30	32
# of electrons	26	26

	Copper: $^{63}_{29}\text{Cu}$	Copper: $^{65}_{29}\text{Cu}$
# of protons	29	29
# of neutrons	34	36
# of electrons	29	29

	Chromium: $^{52}_{24}\text{Cr}$	Chromium: $^{54}_{24}\text{Cr}$
# of protons	24	24
# of neutrons	28	30
# of electrons	24	24

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VM Jensen



Isotope Practice Set:

Embark on a breathtaking journey through nature and adventure with is mesmerizing ebook, Witness the Wonders in **Isotope Practice Set** . This immersive experience, available for download in a PDF format (Download in PDF: *), transports you to the heart of natural marvels and thrilling escapades. Download now and let the adventure begin!

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