

Name Mel

Locker # _____

Bonding & Molecular Structure**LEWIS STRUCTURES**

Indicate the total # of VALENCE electrons for each molecule (VE's). Calculate the number of bonded electrons from the skeleton (BE's). Calculate the Remaining Electrons (RE's = VE's - BE's) and distribute the remaining electrons onto the skeletal structure. Finally, satisfy the octets by creating double and triple bonds. Your final structures should be given below and your scratch paper should clearly show your work.

F_2 VE's = <u>14</u> BE's = <u>2</u> RE's = <u>12</u> $\text{:}\ddot{\text{F}}-\ddot{\text{F}}\text{:}$	H_2 VE's = <u>2</u> BE's = <u>2</u> RE's = <u>0</u> $\text{H}-\text{H}$	HF VE's = <u>8</u> BE's = <u>2</u> RE's = <u>6</u> $\text{H}-\ddot{\text{F}}\text{:}$	NH_3 VE's = <u>8</u> BE's = <u>6</u> RE's = <u>2</u> $\begin{array}{c} \text{H} \\ \\ \text{H}-\text{N}-\text{H} \\ \\ \cdot \end{array}$
Shape: <u>linear</u>	Shape: <u>linear</u>	Shape: <u>linear</u>	Shape: <u>trigonal pyramidal</u>
CH_4 VE's = <u>8</u> BE's = <u>8</u> RE's = <u>0</u> $\begin{array}{c} \text{H} \\ \\ \text{H}-\text{C}-\text{H} \\ \\ \text{H} \end{array}$	C_2H_6 VE's = <u>14</u> BE's = <u>14</u> RE's = <u>0</u> $\begin{array}{c} \text{H} \quad \text{H} \\ \quad \\ \text{H}-\text{C}-\text{C}-\text{H} \\ \quad \\ \text{H} \quad \text{H} \end{array}$	AlH_3 VE's = <u>6</u> BE's = <u>6</u> RE's = <u>0</u> $\begin{array}{c} \text{H}-\text{Al}-\text{H} \\ \\ \text{H} \end{array}$	BH_3 VE's = <u>6</u> BE's = <u>6</u> RE's = <u>0</u> $\begin{array}{c} \text{H} \\ \\ \text{H}-\text{B}-\text{H} \end{array}$
Shape: <u>tetrahedral</u>	Shape: <u>tetrahedral (both)</u>	Shape: <u>trigonal planar</u>	Shape: <u>trigonal planar</u>
C_2H_4 VE's = <u>12</u> BE's = <u>12</u> RE's = <u>0</u> $\begin{array}{c} \text{H} \quad \text{H} \\ \diagdown \quad / \\ \text{C}=\text{C} \\ / \quad \diagdown \\ \text{H} \quad \text{H} \end{array}$	C_2F_4 VE's = <u>36</u> BE's = <u>12</u> RE's = <u>24</u> $\begin{array}{c} \text{:}\ddot{\text{F}}\text{:} \quad \text{:}\ddot{\text{F}}\text{:} \\ \diagdown \quad / \\ \text{C}=\text{C} \\ / \quad \diagdown \\ \text{:}\ddot{\text{F}}\text{:} \quad \text{:}\ddot{\text{F}}\text{:} \end{array}$	CO VE's = <u>10</u> BE's = <u>6</u> RE's = <u>4</u> $\text{:}\ddot{\text{C}}\equiv\ddot{\text{O}}\text{:}$	O_2 VE's = <u>12</u> BE's = <u>4</u> RE's = <u>8</u> $\text{:}\ddot{\text{O}}=\ddot{\text{O}}\text{:}$
Shape around one C: <u>trigonal planar</u>	Shape around one C: <u>trigonal planar</u>	Shape: <u>linear</u>	Shape: <u>linear</u>

Lewis Dot Structures Worksheet Answer Key

Steve Lewis



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Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998

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Using ICT to Enhance Teaching and Learning in Chemistry Steve Lewis, 2004 Information and Communications Technology has revolutionised the ways we process access and use information and as computers and other devices become ever more powerful and information becomes more readily available the next generation will need to be able to interact with digital media effectively to exploit these amazing new technologies to their full potential for the benefit of society It is therefore imperative that teachers become familiar with ICT and its true potential and can present information with a perspective similar to that which the present generation of young people is using to

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High Leverage Practices for Inclusive Classrooms James McLeskey, Lawrence Maheady, Bonnie Billingsley, Mary T. Brownell, Timothy J. Lewis, 2022-03-30 High Leverage Practices for Inclusive Classrooms Second Edition offers a set of practices that are integral to the support of student learning and that can be systematically taught learned and implemented by those entering the teaching profession In this second edition chapters have been fully updated to reflect changes in the field since its original publication and feature all new examples illustrating the use of HLPs and incorporating culturally responsive practices Focused primarily on Tiers 1 and 2 or work that mostly occurs with students with mild to moderate disabilities in general education classrooms this powerful research based resource provides rich practical information highly suitable for teachers and additionally useful for teacher educators and teacher preparation programs

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