



## Student Exploration: Chemical Equations

**Essential Question:** How do the number of atoms of each element in a chemical equation compare to the number of atoms of each element in the products? How do the number of molecules of each element in a chemical equation compare to the number of molecules of each element in the products? Why?

**What You Will Do:** Investigate how the number of atoms of each element in a chemical equation compares to the number of atoms of each element in the products.

1. Analyze the chemical equation for the reaction of hydrogen and oxygen to form water. How many molecules of each element are in the reactants? How many molecules of each element are in the products? How do the number of atoms of each element in the reactants compare to the number of atoms of each element in the products? How do the number of molecules of each element in the reactants compare to the number of molecules of each element in the products? Why?
2. Repeat the investigation for the reaction of hydrogen and oxygen to form hydrogen peroxide. How do the number of atoms of each element in the reactants compare to the number of atoms of each element in the products? How do the number of molecules of each element in the reactants compare to the number of molecules of each element in the products? Why?

### What You Will Do

Hydrogen is an extremely flammable gas. It reacts with oxygen to form water. The chemical equation for this reaction is shown below. The number of atoms of each element in the reactants and products is shown in the table below.

Hydrogen peroxide is a chemical compound that is made up of hydrogen and oxygen atoms. The chemical equation for the reaction of hydrogen and oxygen to form hydrogen peroxide is shown below. The number of atoms of each element in the reactants and products is shown in the table below.

Reactants	Products
2H <sub>2</sub> + O <sub>2</sub>	2H <sub>2</sub> O

1. Analyze the chemical equation for the reaction of hydrogen and oxygen to form water.
  - a. How many hydrogen atoms are in the reactants? 4 products
  - b. How many oxygen atoms are in the reactants? 2 products



# Explore Learning Chemical Equations Answers

**E Durkheim**



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