

## Gas Laws Simulation

Use the following simulation to explore the relationship between pressure, volume, and temperature.

1. Describe the three different gases (pressure, molecular mass, and shape).



2. For each gas, describe the relationship between pressure and volume. What is the relationship? (What happens?)
3. How do pressure? Particles behave differently when volume and pressure (What happens?)
4. What the relationship is for a fixed amount of gas in the simulation?
5. How do pressure? Particles are very dependent and moving randomly along the container (What happens when the simulation?)

Investigate the relationship:

1. Investigate the relationship between the volume and temperature of the gas.
2. What the relationship is?
3. How do the molecules behave when the volume and temperature?
4. What the relationship is?
5. How do the molecules behave when the volume and temperature?

Temperature, pressure, volume	
Temperature (K)	Volume (m)
200	1.00
300	1.50
400	2.00

6. How do pressure, volume, and temperature of a gas relate to the pressure of a gas? How do the temperature of a quantity of the gas in a container, the pressure of a gas in a container, influence temperature?

# Exploring The Gas Laws Simulation Part Answers

**RJ Alexander**



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