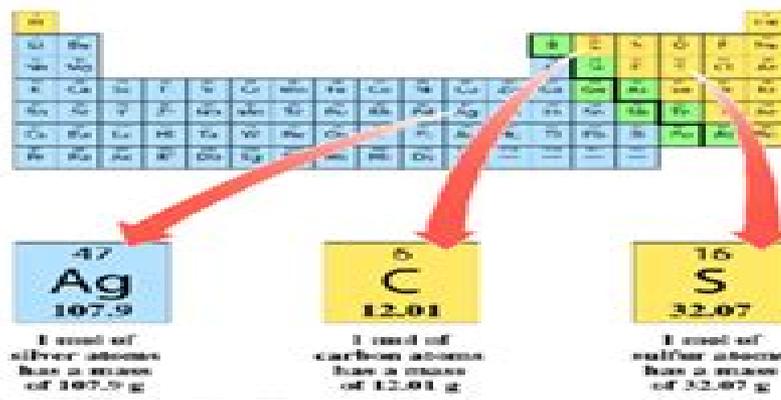


## Chapter 7 – Chemical Quantities

### Molar Mass

- We can't measure the mass of one atom! Too \_\_\_\_\_
- Instead, we work with \_\_\_\_\_ of atoms or molecules.
  - Since it is a convenient size for the lab ("handful amounts")
- Determine molar mass by looking at the \_\_\_\_\_
  - The number \_\_\_\_\_ the element symbol is the mass of one mole of the element with the unit \_\_\_\_\_



- Look at your Periodic Table (inside cover of textbook) and write the molar masses of the following elements. Always record all the digits that the periodic table gives you!

Cu \_\_\_\_\_ g

Ca \_\_\_\_\_ g

O \_\_\_\_\_ g

Pop up question 1

# Guided Notes Chapter Chemical Quantities

**IM Harris**



## **Guided Notes Chapter Chemical Quantities:**

## Unveiling the Energy of Verbal Art: An Psychological Sojourn through **Guided Notes Chapter Chemical Quantities**

In some sort of inundated with monitors and the cacophony of instantaneous communication, the profound energy and mental resonance of verbal art often diminish in to obscurity, eclipsed by the constant assault of noise and distractions. However, located within the lyrical pages of **Guided Notes Chapter Chemical Quantities**, a charming function of fictional elegance that pulses with fresh feelings, lies an wonderful journey waiting to be embarked upon. Published by a virtuoso wordsmith, that enchanting opus manuals readers on a mental odyssey, lightly exposing the latent possible and profound impact embedded within the intricate web of language. Within the heart-wrenching expanse with this evocative examination, we will embark upon an introspective exploration of the book is key themes, dissect its charming writing type, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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