

Matter Test Study Guide



matter = anything that has mass and takes up space.

pure substance = a single kind of matter that is pure, meaning that it always has a specific chemical makeup. Ex.) table salt is always NaCl and sugar is always C₁₂H₂₂O₁₁

element= a pure substance that can't be broken down into any other substance. Ex.) oxygen, copper, aluminum, zinc, tungsten, carbon, hydrogen, helium, nitrogen

atom= the smallest part of an element. Ex.) one atom of oxygen



compound= a pure substance made of two or more elements chemically combined in a set ratio. Ex.) H₂O

molecule= the smallest part of a compound Ex.) one molecule of H₂O



mixture = made of two or more substances that are mixed together, but not chemically combined. Ex.) trail mix

homogeneous= can't see the different parts. Ex.) sugar water

heterogeneous= can see the different parts. Ex.) salad



solution = an example of a homogenous mixture

chemical formulas= tell you which elements and how many atoms of each are in a compound ex.) C₁₂H₂₂O₁₁

set ratio = a specific number of atoms of each element to make a compound

weight = a measure of the force of gravity on an object; can change- depends on location
Ex.) earth vs. moon

mass = a measure of the amount of matter in an object; **does not change**

- triple beam balance
- SI unit is the kilogram
- scientists prefer to measure by mass rather than weight because mass doesn't change



My **WEIGHT** on Earth is around 560N



My **WEIGHT** on the moon is around 90N



My **MASS** is always 56kg!

Matter Study Guide

Ying-Ying Zheng



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