

DAT General Chemistry practice test for 2024 candidates with best marking scheme

1. Atomic Number - answers>number of protons found in that atom; low number; (Z)
2. Mass Number - answers>Protons + Neutrons; high number; (A)
3. Atomic Mass - answers>relative mass of that atom compared to the mass of a carbon-12 atom (which is set at 12.00); measured in amu;
4. 1 amu = - answers> 1.66×10^{-24}
5. Atomic Weight - answers>weight (g) of one mole (mol) of a given element (g/mol); the weighted average of all the masses (not the weights) of an element, weighted according to the natural abundance of the isotopic species of an element
6. 1mol = _____ (Avagadro's Number) - answers> 6.022×10^{23} molecules
7. Isotope - answers>multiple species of an atom with same number of protons and a different number of neutrons (different mass numbers); generally exhibit same chemical properties
8. Ernest Rutherford - answers>1911; gave experimental evidence that an atom has a dense, positively charged

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