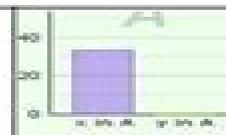


Activity A:
Temperature and diffusion

Get the Gizmo ready:

- Click **Reset** (↺).
- Set the **Wall** to 100%.



Introduction: In this Gizmo, temperature is measured on the **Kelvin scale**. On this scale, 0 K represents **absolute zero**, the coldest possible temperature. Water freezes at 273.15 K (0 °C), and water boils at 373.15 K (100 °C).

Question: How does temperature affect the rate of diffusion?

1. **Observe:** Click **Reset**. Set the temperature to 100 K, and click **Play**. Observe the number of particles that pass through the wall. Then set the temperature to 200 K, and **Play** again.
- How does the temperature of the gas relate to the number of particles that pass through the wall?

As the temperature of the gas increases, the number of particles that pass through the wall increases.

The number of particles that pass through the wall is a measure of the average **kinetic energy** of the particles. Kinetic energy is the energy of motion. Kinetic energy is measured in joules (J).

2. **Observe:** Click **Reset**. Set the temperature to 100 K, and click **Play**. Observe the rate of diffusion.

How does the rate of diffusion relate to the temperature of the gas?

3. **Observe:** Click **Reset**. Set the temperature to 100 K, and click **Play**. Set the wall to 100%. Observe the rate of diffusion. Then click **Play** again. Observe the rate of diffusion. Then click **Play** again. Observe the rate of diffusion.

How does the rate of diffusion relate to the temperature of the gas? How does the rate of diffusion relate to the wall's permeability?

Report your observations to 100 K, and then set the wall to 50%. Observe the rate of diffusion.

Temp = 100 K, permeability = 50%

Temp	100 K	Permeability	50%
Rate of diffusion			

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EW Minium



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