



WINTER- 18 EXAMINATION

Subject Name: Data Structure using C

Model Answer

Subject Code:

22317

Important Instructions to examiners:

- 1) The answers should be examined by key words and not as word-to-word as given in the model answer scheme.
- 2) The model answer and the answer written by candidate may vary but the examiner may try to assess the understanding level of the candidate.
- 3) The language errors such as grammatical, spelling errors should not be given more importance (Not applicable for subject English and Communication Skills).
- 4) While assessing figures, examiner may give credit for principal components indicated in the figure. The figures drawn by candidate and model answer may vary. The examiner may give credit for any equivalent figure drawn.
- 5) Credits may be given step wise for numerical problems. In some cases, the assumed constant values may vary and there may be some difference in the candidate's answers and model answer.
- 6) In case of some questions credit may be given by judgement on part of examiner of relevant answer based on candidate's understanding.
- 7) For programming language papers, credit may be given to any other program based on equivalent concept.

Q. No.	Sub Q. N.	Answer	Marking Scheme
1		Attempt any FIVE of the following :	10 M
	a	Define the term algorithm.	2 M
	Ans	Algorithm is a stepwise set of instructions written to perform a specific task.	Correct definition 2M
	b	List any 4 applications of queue.	2 M
	Ans	<ul style="list-style-type: none"><li>• In computer system to maintain waiting list for single shared resources such as printer, disk, etc.</li><li>• It is used as buffers on MP3 players, iPod playlist, etc.</li><li>• Used for CPU scheduling in multiprogramming and time sharing systems.</li><li>• In real life, Call Center phone systems will use Queues, to hold people calling them in an order, until a service representative is free.</li></ul>	Any four applications- 1/2 M each

# Msbte Model Answer Paper 2for 3rd Sem

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web the balanced chemical equation for this reaction is  $\text{Mg}(\text{s}) + 2\text{HNO}_3(\text{aq}) \rightarrow \text{Mg}(\text{NO}_3)_2(\text{aq}) + \text{H}_2(\text{g})$  in this equation the s symbolizes solid magnesium aq represents aqueous nitric acid aq signifies magnesium nitrate in a dissolved state and g represents the gaseous hydrogen produced

$\text{HNO}_3 + \text{Mg}(\text{C}_2\text{H}_3\text{O}_2)_2 \rightarrow \text{Mg}(\text{NO}_3)_2 + \text{C}_2\text{H}_3\text{O}_2$  chemical equation - Aug 11 2023

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aqueous magnesium nitrate  $\text{Mg(NO}_3)_2$  and one mole of dihydrogen  $\text{H}_2$  gas show chemical

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web magnesium carbonate nitric acid magnesium nitrate water carbon dioxide one mole of magnesium carbonate  $\text{MgCO}_3$  and two moles of nitric acid  $\text{HNO}_3$  react to form one mole of magnesium nitrate  $\text{Mg(NO}_3)_2$  one mole of water  $\text{H}_2\text{O}$  and one mole of carbon dioxide  $\text{CO}_2$  show chemical structure image reactants

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web nov 20 2020 in this video we ll balance the equation  $\text{mg hno}_3 \text{ mg no}_3 \text{ 2 no}_2 \text{ h}_2\text{o}$  and provide the correct coefficients for each compound to balance  $\text{mg hno}_3 \text{ mg no}_3 \text{ 2 no}_2 \text{ h}_2\text{o}$  you ll need to be

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web magnesium hydroxide nitric acid balanced total equation  $\text{mg oh}_2 \text{ aq 2 hno}_3 \text{ aq 2 h}_2\text{o mg no}_3 \text{ 2 aq}$  total ionic equation  $\text{mg}_2 \text{ 2 oh}_1 \text{ 2 h}_1 \text{ 2no}_3 \text{ 1 2h}_2 \text{ o mg}_2 \text{ 2 no}_3 \text{ 1}$  net ionic equation  $\text{2 oh}_1 \text{ 2 h}_1 \text{ 2h}_2 \text{ o}$  sulfuric acid and lithium hydroxide balanced total equation  $\text{h}_2 \text{ so}_4 \text{ aq 2 lihoh aq 2 h}_2$