

**RELIANCE INTERNATIONAL ACADEMY**

Saraswatinagar, Kathmandu

**Model set 3**Grade: - 10  
Subject: - Chemistry  
Stream: - Science**SET-3**Full Marks: - 75  
Pass Marks: - 30  
Time: - 3 hrs.

Candidates are required to give answers in their own words as far as practicable. Figures in the margin indicate full marks. Do not scatter the answers.

**Group – 'A'****Multiple choice questions**

Attempt all questions.

**[11 × 1 = 11]**

1.  $PCl_5$ ,  $PH_3$  and  $HCl$  follow
- Law of conservation of mass.
  - law of constant proportion.
  - law of multiple proportion.
  - law of reciprocal proportion.
2. Number of atoms present in 0.1 mole of water is
- 0.1N<sub>A</sub>
  - 0.2N<sub>A</sub>
  - 0.3N<sub>A</sub>
  - 0.4N<sub>A</sub>
3. Shape of ammonia is
- pyramidal
  - Tetrahedral
  - coplanar
  - Trigonal

4. Essential component of amalgam is

- Fe
- Zn
- Cu
- Hg

5. Petroleum is source of

- Aliphatic hydrocarbon
- Aromatic hydrocarbon
- Alicyclic hydrocarbon
- Heterocyclic hydrocarbon

6. The compound formed in the detection of nitrogen in organic compound is

- $Fe_4[Fe(CN)_6]_3$
- $Fe_3[Fe(CN)_6]_2$
- $Fe_2[Fe(CN)_6]$
- $Fe_4[Fe(CN)_6]_3$

7. Which of the following are recycled in the manufacture of sodium carbonate by solvay's process

- $CO_2$  and  $NH_4Cl$
- $CO_2$  and  $NH_3$
- $NaCl$  and  $CaO$
- $NaCl$  and  $NH_3$

8. oxidation number of Cr in  $K_2Cr_2O_7$  is

- +1
- +2
- +4
- +6

9. Which of the following is aqua fortis?

- $HNO_3$
- $HCl$
- $H_3PO_4$
- $H_2SO_4$

10. Electronegativity of elements is in the order

- $F > Cl > Br > I$
- $I > Br > Cl > F$
- $Cl > F > Br > I$
- $Br > I > Cl > F$

11. which of the following reagent can be used to detect sulphate ion?

- $NaOH$
- $BaCl_2$
- Conc.  $HNO_3$
- $AgNO_3$

**Short answer questions**

Attempt All the questions

**[5 × 4 = 20]**

1. An element X has 2 electrons in K shell, 8 electrons in L shell and 5 electrons in M shell.

- Identify the element X and write the number of protons and electrons in it.
- Size of X ion is greater than that of X atom though both contain the same number of protons. Give reason.
- Write down the formula and name of one of the compounds formed by X. **[3+1+1]**

OR

Give a balanced chemical reaction for the preparation of  $HCl$  and  $HBr$  in the laboratory. How would you test the presence of  $Cl^-$ ,  $Br^-$  and  $I^-$  ion's in it's aqueous state

**[2+1+2]**

2. Balance the following reaction by oxidation number method or ion electron method. Identify oxidant and reductant in the following reaction.



3. State and explain Le-Chatelier's principle. Derive the relationship between  $K_p$  and  $K_c$ . **[2+4]**

OR

The kinetic theory of gases explains the macroscopic properties of gases, such as volume, pressure, and temperature as well as viscosity, thermal conductivity and mass diffusivity. The model also accounts for related phenomena such as Brownian motion.

- Write the main postulates of kinetic theory of gas. **[2]**
- What is absolute zero temperature? **[1]**
- Calculate the number of moles of molecules present in 30 ml of an ideal gas exerting a pressure of 770 mm at  $25^\circ C$ ? **[1]**

4. Ammonia gas is manufactured by Haber's process.

- Give the principle chemical reaction **[1]**
- Write the necessary conditions for the reaction **[2]**

# L O Question Paper 2013 Grade11

**Huangqi Zhang**



**L O Question Paper 2013 Grade11:**

## **L O Question Paper 2013 Grade11** Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the energy of words has been evident than ever. They have the capability to inspire, provoke, and ignite change. Such is the essence of the book **L O Question Paper 2013 Grade11**, a literary masterpiece that delves deep into the significance of words and their effect on our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book's key themes, examine its writing style, and analyze its overall impact on readers.

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### **Table of Contents L O Question Paper 2013 Grade11**

1. Understanding the eBook L O Question Paper 2013 Grade11
  - The Rise of Digital Reading L O Question Paper 2013 Grade11
  - Advantages of eBooks Over Traditional Books
2. Identifying L O Question Paper 2013 Grade11
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an L O Question Paper 2013 Grade11
  - User-Friendly Interface
4. Exploring eBook Recommendations from L O Question Paper 2013 Grade11
  - Personalized Recommendations
  - L O Question Paper 2013 Grade11 User Reviews and Ratings
  - L O Question Paper 2013 Grade11 and Bestseller Lists
5. Accessing L O Question Paper 2013 Grade11 Free and Paid eBooks

- L O Question Paper 2013 Grade11 Public Domain eBooks
  - L O Question Paper 2013 Grade11 eBook Subscription Services
  - L O Question Paper 2013 Grade11 Budget-Friendly Options
6. Navigating L O Question Paper 2013 Grade11 eBook Formats
- ePub, PDF, MOBI, and More
  - L O Question Paper 2013 Grade11 Compatibility with Devices
  - L O Question Paper 2013 Grade11 Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of L O Question Paper 2013 Grade11
  - Highlighting and Note-Taking L O Question Paper 2013 Grade11
  - Interactive Elements L O Question Paper 2013 Grade11
8. Staying Engaged with L O Question Paper 2013 Grade11
- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers L O Question Paper 2013 Grade11
9. Balancing eBooks and Physical Books L O Question Paper 2013 Grade11
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection L O Question Paper 2013 Grade11
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine L O Question Paper 2013 Grade11
- Setting Reading Goals L O Question Paper 2013 Grade11
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of L O Question Paper 2013 Grade11
- Fact-Checking eBook Content of L O Question Paper 2013 Grade11
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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