

Study Guide for Chemistry EOC Assessment

Your test will consist of multiple-choice questions over the following topics.

1. Metric units for mass, length, volume
2. Metric to metric conversions (K H Da ml/g D C M)
3. Know how to put numbers in scientific notation and take them out.
4. Significant Digit rules
5. Difference between chemical and physical properties and changes. There will be multiple questions on this.
6. Ways to tell if a chemical change (chemical reaction) is taking place.
7. Pure substances and mixtures
 - a. heterogeneous mixture
 - b. homogeneous mixture
8. Density & how it relates to floating and sinking
9. Use the equation for density
$$\text{Density} = \frac{\text{Mass}}{\text{volume}}$$

(g/ml) (g) (ml)
10. Know the phases of matter and the characteristics of each phase.
11. Law of Conservation of Matter
12. What is a mole in chemistry?
13. Convert grams to moles, and moles to grams
14. Define an atom
15. Structure of an atom and where each subatomic particle is located.
 - a. Nucleus (contains protons and neutrons)
 - b. Protons (positive)
 - c. Neutrons (neutral)
 - d. Electrons (negative)
16. Isotopes
17. Elements
18. If given atomic # and mass, find the number of protons and neutrons in an atom.
19. Find the number of valence electrons, and tell what it says about reactivity.
20. Periodic Table
 - a. Know about the families
 - b. Group vs. period
21. Know about the properties of metals and nonmetals.
 - a. Ductile, malleable, conductors
22. Ionic & Covalent Compounds
How do they form? (Transfer or gain electrons)
Characteristics of Each
23. Be able to name ionic compounds
24. Predict oxidation numbers
25. Chemical Equations:
Know the parts of an equation
5 main types of reactions-know about them and be able to identify them
(Synthesis, Decomposition, Single Displacement, Double Displacement, Combustion)
26. Acids & Bases-
Macro and Microscopic properties of each
What ions do they form, how do they affect litmus, what is their pH range?
Identify examples of each- Mono, di, and polyprotic acids
Tell if they are strong or weak
Acid-base indicator

Eoc Chemistry Study Guide

D Keegan



Eoc Chemistry Study Guide:

Reviewing **Eoc Chemistry Study Guide**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Eoc Chemistry Study Guide**," an enthralling opus penned by a highly acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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