

## Stoichiometry Sample Lesson Plan

Descriptive Data:	Kris Thompson Unit: Conservation of Matter 10th Grade Chemistry
Lesson Name	Unit Assessment - Story Starter
State Standards:	<b>HSPS1-7</b> Use mathematical representation to support the claim that atoms and therefore mass, are conserved during a chemical reaction. <b>PS1B</b> The fact that atoms are conserved together with knowledge of chemical properties of the elements involved can be used to describe and predict chemical reactions.
Measurable Objectives to be assessed:	<ul style="list-style-type: none"> <li>• Students apply the Law of conservation of Matter to balance chemical equations.</li> <li>• Students apply the proportional relationships between masses of atoms on a macroscopic scale to determine the amount of substance consumed or produced in a chemical reaction</li> <li>• Students identify chemical equations based on the amount of substance produced or consumed</li> <li>• Students explain how matter is conserved in a chemical reaction</li> </ul>
Materials	Baking Soda (2g per group) Test tubes (2 per group) Test tube holder Electronic Balance (3 total) Bunsen Burner (1 per group) spatulas (2 per group) weighing dishes spoon periodic table calculators (at least 1 per group) Absent Minded Professor Graphic Organizer (See Below) <a href="https://www.youtube.com/watch?v=mMvYr7XCg_5Y">https://www.youtube.com/watch?v=mMvYr7XCg_5Y</a>
Safety	Students must adhere to safety precautions when working with a Bunsen burner: Know where all fire extinguishers are in the classroom Do not place any hot objects on top of the electronic balance Use caution when handling objects that have been heated Lab Coat and Safety Goggles must be worn during this experiment Sodium Bicarbonate MSDS: <a href="http://www.sciencelab.com/msds.php?msdsid=9927258">http://www.sciencelab.com/msds.php?msdsid=9927258</a>
Engage (5-10 min)	Students are introduced to why baking soda is used in baking and the decomposition process that occurs by reading the background information about baking soda.
Explore (35-40 min)	Pair up students to complete this portion of the assessment. Students use the designated materials to perform a decomposition reaction of $\text{NaHCO}_3$ . Procedure <ol style="list-style-type: none"> <li>1. Measure and record the mass of all your materials.</li> <li>2. Place an empty weigh boat on the balance pan and then press the tare/reset button.</li> </ol>

# Lesson Plan Stoichiometry

**Rachel Sandford**



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