

## Chemistry Worksheets Class 8 on Chapter 4 Materials Metals and Non metals - Set 1

**Q-1:** Which of the following is not a characteristic of metals?

- a) Can be drawn into wires
- b) Generally hard
- c) produces stillness when dropped
- d) Shiny appearance

**Q-2:** Which of the following statements is not correct?

- a) Iron is a metal and it reacts with water vigorously
- b) Reactions of non metals with base are complex
- c) Non metallic oxides are acidic and metallic oxides are basic
- d) Sodium metal cannot be kept in the open air

**Q-3:** Which of the following metal/non metal is stored in water?

- a) Na
- b) K
- c) S
- d) P

**Q-4:** In the table below, identify the metal or nonmetal present in them that forms oxides with oxygen, as well as the chemical formula of the acids and bases from their corresponding nonmetal or metal.

S.No	Chemical formula of the acid	Non metal	Chemical formula of the base	Metal
1.	$H_3PO_4$		$Na_2O$	
2.		Carbon		Potassium
3.	$H_2SO_4$		$NH_4OH$	
4.	HCl			Magnesium
5.		Nitrogen	$Fe_2O_3$	

**Q-5:** Fill in the blanks

- a) The metal which is the best conductor of heat and electricity is \_\_\_\_\_.
- b) Out of tin, lead, copper and mercury, the metals that lie above hydrogen in the activity series are \_\_\_\_\_ and \_\_\_\_\_.

# Metals Section Ch 19 Answers

**Rex Miller, Mark R. Miller**



## Metals Section Ch 19 Answers:

**Metal Ions and Complexes in Solution** Toshio Yamaguchi, Ingmar Persson, 2023-12-04 Based on a translated Japanese title published in 2012 this book provides fundamental aspects of experimental and computational methods the properties and structure of solvents ion solvation and equilibria and reactions of metal complexes in solution It includes state of the art details on metal complexes in newly developing sustainable liquids and applications in real life Appealing to researchers working in coordination chemistry including students and industrialists the text uses exercises tables and figures to help the reader with their understanding of the topic Descriptive Inorganic Chemistry Student's Solutions Manual Geoff

Rayner-Canham, Tina Overton, 2005-12-21 Solutions for all odd numbered problems in text Chemistry for the Utterly Confused John T. Moore, Richard H. Langley, 2007-06-01 Banish bafflement in this tough subject From formulas and lab techniques to the periodic table Chemistry for the Utterly Confused focuses on the areas of maximum confusion and breaks down the most difficult chemistry topics into easy to understand concepts This invaluable guide also teaches problem solving skills you need to master this imposing subject Whether you re in high school in college or simply brushing up on chemistry knowledge this fun easily accessible book will make understanding chemistry a breeze *Chemistry* John A.

Olmsted, Gregory M. Williams, Robert Charles Burk, 2016-01-14 Olmsted Burk is an introductory general chemistry text designed specifically with Canadian professors and students in mind A reorganized Table of Contents and inclusion of SI units IUPAC standards and Canadian content designed to engage and motivate readers distinguish this text from many of the current text offerings It more accurately reflects the curriculum of most Canadian institutions Instructors will find the text sufficiently rigorous while it engages and retains student interest through its accessible language and clear problem solving program without an excess of material that makes most text appear daunting and redundant **Federal Register**, 1951-10

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