

APPENDIX G

Standard Thermodynamic Properties for Selected Substances

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Substance	ΔH_f° (kJ mol ⁻¹)	ΔG_f° (kJ mol ⁻¹)	S° (J K ⁻¹ mol ⁻¹)
aluminum			
Al(s)	0	0	28.3
Al(g)	324.4	285.7	164.54
Al ³⁺ (aq)	-531	-485	-321.7
Al ₂ O ₃ (s)	-1676	-1582	50.92
AlF ₃ (s)	-1510.4	-1425	66.5
AlCl ₃ (s)	-704.2	-628.8	110.67
AlCl ₃ ·6H ₂ O(s)	-2691.57	-2269.40	376.56
Al ₂ S ₃ (s)	-724.0	-492.4	116.9
Al ₂ (SO ₄) ₃ (s)	-3445.06	-3506.61	239.32
antimony			
Sb(s)	0	0	45.69
Sb(g)	262.34	222.17	180.16
Sb ₂ O ₃ (s)	-1440.55	-1268.17	220.92
SbCl ₃ (g)	-313.8	-301.2	337.80
SbCl ₃ (l)	-394.34	-334.29	401.94
Sb ₂ S ₃ (s)	-174.89	-173.64	182.00
SbCl ₅ (s)	-382.17	-323.72	184.10
SbOCl(s)	-374.0	—	—

TABLE G.1

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Modern Chemistry Appendix D:

Unveiling the Energy of Verbal Art: An Mental Sojourn through **Modern Chemistry Appendix D**

In a global inundated with displays and the cacophony of fast communication, the profound energy and psychological resonance of verbal beauty often diminish in to obscurity, eclipsed by the regular onslaught of noise and distractions. However, located within the lyrical pages of **Modern Chemistry Appendix D**, a charming function of fictional beauty that pulses with raw thoughts, lies an wonderful journey waiting to be embarked upon. Published by a virtuoso wordsmith, that mesmerizing opus manuals readers on an emotional odyssey, delicately exposing the latent potential and profound impact stuck within the elaborate web of language. Within the heart-wrenching expanse of this evocative examination, we will embark upon an introspective exploration of the book is main subjects, dissect their fascinating writing type, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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