

Name: \_\_\_\_\_ ( ) Class: \_\_\_ Date: \_\_\_\_\_

## A. Multiple choice questions (10 marks)

1. Which of the following statements concerning vinegar is/are correct?

- (1) The process of forming hydrogen ions in vinegar is reversible.  
(2) Neutralization occurs when sugar is added to vinegar.  
(3) The pH of vinegar used in kitchen is around 1.

- A. (1) only  
B. (2) only  
C. (1) and (2) only  
D. (2) and (3) only

2. 25.00 cm<sup>3</sup> of 0.50 M lead(II) nitrate solution is mixed with 50.00 cm<sup>3</sup> of 1.00 M sodium chloride solution. Insoluble lead(II) chloride is formed during mixing. What is the concentration of Cl<sup>-</sup>(aq) in the mixture?

- A. 0.33 M  
B. 0.50 M  
C. 0.75 M  
D. 1.50 M

3. Consider the solutions W, X, Y, Z below:

W	100 cm <sup>3</sup> of 0.20 M HNO <sub>3</sub> (aq)
X	50 cm <sup>3</sup> of 0.20 M HCl (aq)
Y	100 cm <sup>3</sup> of 0.20 M CH <sub>3</sub> CO <sub>2</sub> H (aq)
Z	50 cm <sup>3</sup> of 0.10 M NaOH (aq)

Which of the following statements is/are correct?

- A. The pH of Y equals to  $-\log 0.2$ .  
B. Mixing W and Z gives a neutral solution.  
C. The pH of the mixture of W and X is lower than that of W.  
D. The pH of the mixture of W and X is lower than that of the mixture of X and Y.

4. Which of the following statements about ethanoic acid is correct?

- A. It is a strong acid.  
B. Its empirical formula is CH<sub>2</sub>O.  
C. It forms a salt NaCH<sub>2</sub>COO when reacting with NaOH (aq).

# Modern Chemistry Ch14 Review Answer Key

**ML Morrison**



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