

CHAPTER 8 REVIEW

Chemical Equations and Reactions

SHORT ANSWER Answer the following questions in the space provided.

2. _____ According to the law of conservation of mass, the total mass of the reacting substances is
- always more than the total mass of the products.
 - always less than the total mass of the products.
 - sometimes more and sometimes less than the total mass of the products.
 - always equal to the total mass of the products.**

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 2. In the reaction described by the word equation

$$\text{sodium} + \text{water} \rightarrow \text{sodium hydroxide} + \text{hydrogen gas}$$
 - sodium hydroxide is a product.
 - sodium is a product.
 - water is a reactant.
 - Both (a) and (c)**
- _____ 3. Which of the following indicates that a chemical equation is balanced?
 - The numbers of atoms of each element are the same on both sides of the equation.**
 - All of the coefficients are the same.
 - The numbers of molecules on each side of the equation are equal.
 - The sums of the coefficients on each side of the equation are equal.
- _____ 4. From a complete and correctly written chemical equation, you can obtain the
 - chemical formulas of the reactants and products.
 - relative amounts of the reactants and products.
 - physical states of the reactants and products.
 - All of the above**
- _____ 5. To balance a chemical equation, you must
 - adjust the subscripts.
 - adjust the number of products.
 - adjust the number of reactants.
 - adjust the coefficients.**
- _____ 6. The formation of a solid product in a chemical reaction is represented by the symbol
 - (aq).
 - (g).
 - (l).
 - (s).**
- _____ 8. In the expression 3CO_2 , the numbers 3 and 2 are, respectively,
 - a subscript and a coefficient.
 - a coefficient and a subscript.**
 - two subscripts.
 - two coefficients.
- _____ 9. What is the total number of molecules that is represented by the following chemical equation? (Add up the coefficients.)

$$\text{NH}_4\text{NO}_2 \rightarrow \text{N}_2 + 2\text{H}_2\text{O}$$
 - 3
 - 4**
 - 8
 - 16
- _____ 10. Which equation below violates the law of conservation of mass?
 - $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 - $\text{KCl} + \text{Br} \rightarrow \text{KBr} + \text{Cl}_2$**
 - $2\text{Fe}_2\text{O}_3 + 3\text{C} \rightarrow 4\text{Fe} + 3\text{CO}_2$
 - $\text{Na}_2\text{CO}_3 + 2\text{HCl} \rightarrow 2\text{NaCl} + \text{H}_2\text{O} + \text{CO}_2$
- _____ 11. Which of the following equations represents the balanced equation for the reaction of iron and oxygen?
 - $2\text{Fe} + \text{O}_2 \rightarrow \text{Fe}_2\text{O}_3$
 - $\text{Fe} + 3\text{O}_2 \rightarrow \text{Fe}_2\text{O}_3$
 - $4\text{Fe} + 3\text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3$**
 - $3\text{Fe} + 3\text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3$
- _____ 12. Which coefficients correctly balance the formula equation below?

$$\text{KClO}_3(s) \rightarrow \text{KCl}(s) + \text{O}_2(g)$$
 - 1, 1, 1
 - 1, 1, 3
 - 2, 2, 3**
 - 2, 1, 1
- _____ 13. Which equation is *not* balanced?
 - $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 - $4\text{H}_2 + 2\text{O}_2 \rightarrow 4\text{H}_2\text{O}$
 - $\text{H}_2 + \text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{H}_2\text{O}$
 - $2\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$**
- _____ 14. In which type of reaction do two or more compounds react to form one product?
 - synthesis (AKA combination)**
 - decomposition
 - combustion
 - single-replacement
- _____ 15. Which of the following is true regarding single-replacement reactions?
 - Only one product is formed.
 - There is only one reagent.
 - One element replaces another in a compound.**

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web so können sie ihren stoffwechsel auf natürliche weise anregen trinken sie viel wasser das kann den stoffwechsel anregen und beim abnehmen helfen auch grüner tee oder kaffee können effektiv den metabolismus ankurbeln essen sie außerdem viele proteine der körper benötigt mehr energie um proteine zu verarbeiten als diese selbst

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web sie beschleunigen den stoffwechsel und verhindern die fettablagerung im körper zudem wirken viele stoffwechsel tabletten auf das sättigungsgefühl ein und hemmen den appetit so werden der transport und die umwandlung von nährstoffen gewährleistet und gleichzeitig weniger kalorien aufgenommen

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web inhaltsübersicht was ist der stoffwechsel die 4 wichtigsten antworten im bezug auf einen zu langsamen stoffwechsel die drei effektivsten möglichkeiten um kalorien zu verbrennen fazit was ist der stoffwechsel es ist nicht ungewöhnlich dass viele menschen ihre gewichtszunahme auf einen langsamen stoffwechsel zurückführen

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web apr 9 2021 je nachdem wie dein stoffwechsel arbeitet kann er entweder der schlüssel für sportliche höchstleistungen sein oder die quelle hartnäckigen Übergewichts du möchtest ohne gym fett loswerden dieser cleverer trainings und ernährungsplan zum abnehmen eignet sich speziell für anfänger

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web oct 5 2022 vier stoffe die den stoffwechsel beschleunigen können sind koffein capsaicin proteine wasser kann man den stoffwechsel anregen der stoffwechsel kann durch regelmäßige sportliche betätigung eine ausreichende flüssigkeitszufuhr stressreduktion sowie bestimmte lebensmittel angeregt werden

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web der stoffwechsel auch metabolismus genannt umfasst alle vorgänge bei denen stoffe in unseren zellen neu aufgebaut abgebaut oder recycelt werden das ganze dient dazu unseren körper in schuss zu halten man unterscheidet dabei grob zwei stoffwechselabteilungen den baustoffwechsel und den energiestoffwechsel

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weil da der anabole stoffwechsel für die synthese von eiweißen verantwortlich ist spielt dieser prozess auch beim muskelaufbau eine entscheidende rolle wenn du deinem körper ausreichend muskeltreibstoff in form von proteinen zur verfügung stellst kannst du effektiv muskeln aufbauen