

## IONIC BONDING

Ionic bonding occurs when a metal transfers one or more electrons to a nonmetal in an effort to obtain a stable octet of electrons. For example, the transfer of an electron from sodium to chlorine can be depicted by a Lewis dot diagram.



Calcium would need two chlorine atoms to get rid of its two valence electrons.



Show the transfer of electrons in the following combinations.

1.  $\text{K} + \text{F}$



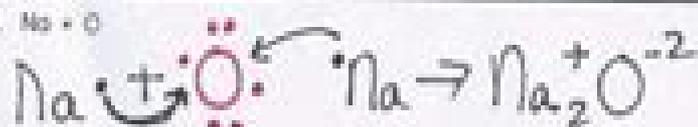
2.  $\text{Mg} + \text{I}$



3.  $\text{Be} + \text{S}$



4.  $\text{Na} + \text{O}$



5.  $\text{Al} + \text{Br}$



## COVALENT BONDING

Name \_\_\_\_\_

Covalent bonding occurs when two or more nonmetals share electrons, attempting to obtain a stable octet of electrons at least part of the time. For example:

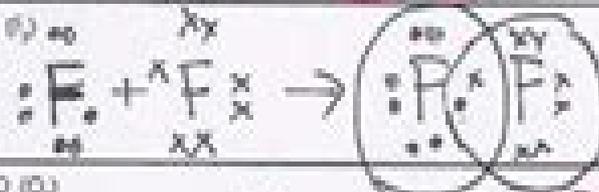


Show how covalent bonding occurs in each of the following pairs of atoms. Atoms may share one, two or three pairs of electrons.

1.  $\text{H} + \text{H} (\text{H}_2)$



2.  $\text{F} + \text{F} (\text{F}_2)$



3.  $\text{O} + \text{O} (\text{O}_2)$



4.  $\text{N} + \text{N} (\text{N}_2)$



5.  $\text{C} + \text{O} (\text{CO}_2)$



6.  $\text{H} + \text{O} (\text{H}_2\text{O})$



# Introduction To Bonding Worksheet Answers Key

**Ensheng Dong**



## Introduction To Bonding Worksheet Answers Key:

Hands-On General Science Activities With Real-Life Applications Pam Walker, Elaine Wood, 2008-04-21 In this second edition of Hands On General Science Activities with Real Life Applications Pam Walker and Elaine Wood have completely revised and updated their must have resource for science teachers of grades 5 12 The book offers a dynamic collection of classroom ready lessons projects and lab activities that encourage students to integrate basic science concepts and skills into everyday life

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*Information Circular*, 1992 The School Science Review, 2006 **The Resource Book** Bryan Strong, 2002 *Fine Homebuilding*, 1995 *Human Sexuality* Strong, 1998-11 National Underwriter, 1950 **The Software Encyclopedia 2000** Bowker Editorial Staff, 2000-05 **Chemical Bonds** Harry B. Gray, 1994-12-05 This profusely illustrated book by a world renowned chemist and award winning chemistry teacher provides science students with an introduction to atomic and molecular structure and bonding This is a reprint of a book first published by Benjamin Cummings 1973 *Chemical Bonding* Audrey L. Companion, 1979 Molecules and the Chemical Bond Henry A. Bent, 2013-09-30 MOLECULES and the Chemical Bond is about understanding Schrdingers equation for chemical systems In his famous Lectures on Physics Richard Feynman quotes Paul Dirac on what it means to understand an equation I understand what an equation means said Dirac if I have a way of figuring out the characteristics of its solutions without actually solving it That hits the nail on the head Its precisely what

Conceptual Valence Bond Theory does for Schrödinger's equation. A physical understanding of an equation adds Feynman is a completely unmathematical imprecise and inexact thing but absolutely necessary for a physicist. It unfolds in MCB in two stages described by Newton as a stage of Analysis a union of observations and inductions and a stage of Synthesis use of inductions accepted as first principles to explain observations. The book's chief vehicle for creating an intuitive understanding of solutions of Schrödinger's equation is the world's largest and to the author's knowledge virtually only library of line drawings of exclusive orbital models of chemical species electron density profiles. By focussing attention on fundamental physical principles and by avoiding use of atomic orbitals and thereby mathematical complexities associated with Schrödinger's equation the only source of atomic orbitals the book's essays provide a scientifically sound student friendly introduction to modern valence theory. Repetition of fundamental ideas here and there is intended to make individual essays understandable and interesting each by itself so that readers may examine them in any order in leisurely walks so to speak in the big garden that is valence theory picking bouquets to their liking.

Atomic Structure and Chemical Bonding Fritz Seel, 1972

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