

1. What volume of hydrogen gas at STP is produced from the reaction of 50.0g of Mg and 75.0 grams of HCl? How much of the excess reagent is left over (in grams)?



Do a standard mass to mass problem starting with each reactant

50.0 g Mg	1 mol Mg 24.31g Mg	1 mol H ₂ 1 mol Mg	22.4 L 1 mol H ₂	= 46.1 L H ₂
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75.0 g HCl	1 mol HCl 36.46 g HCl	1 mole H ₂ 2 mol HCl	22.4 L 1 mol H ₂	= 23.0 L H ₂
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HCl is the limiting reactant!!

Limiting Reactants Pogil Key

Rachel S Tattersall



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Fuel your quest for knowledge with this thought-provoking masterpiece, Dive into the World of **Limiting Reactants Pogil Key**. This educational ebook, conveniently sized in PDF (*), is a gateway to personal growth and intellectual stimulation. Immerse yourself in the enriching content curated to cater to every eager mind. Download now and embark on a learning journey that promises to expand your horizons. .

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