

Chemistry Study Guide

Periodic Table of the Elements

1	2											3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18																																																																								
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K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe	Cs	Ba											Rn																																																			
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Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles 🎯

Matter:

Anything that has **mass** and **occupies space** ✨🌍

Examples: air, gases, liquids, solidz!

Atoms:

Atoms are the **smallest, indivisible particle** 🔬👉

- **Outer part** → *shell/orbit/energy level* → carries the electrons (e^-)
- **Inner part** → *nucleus* → houses the protons (p^+) and neutrons (n)
- Outer & Inner part → together these are the **subatomic particles**

Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	p^+	1	+1	Nucleus
Neutron	n	1	0	Nucleus
Electron	e^-	~ 0	-1	Orbit/Shell

Intro To Chemistry Study Guide

Charles Richmond Henderson



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Table of Contents Intro To Chemistry Study Guide

1. Understanding the eBook Intro To Chemistry Study Guide
 - The Rise of Digital Reading Intro To Chemistry Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Intro To Chemistry Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Intro To Chemistry Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Intro To Chemistry Study Guide
 - Personalized Recommendations
 - Intro To Chemistry Study Guide User Reviews and Ratings
 - Intro To Chemistry Study Guide and Bestseller Lists
5. Accessing Intro To Chemistry Study Guide Free and Paid eBooks
 - Intro To Chemistry Study Guide Public Domain eBooks
 - Intro To Chemistry Study Guide eBook Subscription Services
 - Intro To Chemistry Study Guide Budget-Friendly Options
6. Navigating Intro To Chemistry Study Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Intro To Chemistry Study Guide Compatibility with Devices
 - Intro To Chemistry Study Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Intro To Chemistry Study Guide
 - Highlighting and Note-Taking Intro To Chemistry Study Guide
 - Interactive Elements Intro To Chemistry Study Guide
8. Staying Engaged with Intro To Chemistry Study Guide

- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Intro To Chemistry Study Guide
9. Balancing eBooks and Physical Books Intro To Chemistry Study Guide
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Intro To Chemistry Study Guide
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Intro To Chemistry Study Guide
- Setting Reading Goals Intro To Chemistry Study Guide
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Intro To Chemistry Study Guide
- Fact-Checking eBook Content of Intro To Chemistry Study Guide
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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