

| Question     | Answer   | Mark       | Guidance  |                |   |  |            |          |            |            |  |      |      |      |      |                |   |  |
|--------------|--|------------|---|----------------|---|--|------------|----------|------------|------------|--|------|------|------|------|----------------|---|--|
| 2 (a)        | <p>FIRST CHECK THE ANSWER ON THE ANSWER LINE<br/>IF answer = <math>\text{CH}_4\text{N}_2\text{O}</math> award 2 marks</p> <table border="0" style="width: 100%; text-align: center;"> <tr> <td>C</td> <td>H</td> <td>N</td> <td>O</td> <td></td> </tr> <tr> <td>20.00/12.0</td> <td>6.67/1.0</td> <td>46.67/14.0</td> <td>26.66/16.0</td> <td></td> </tr> </table> <p>OR</p> <table border="0" style="width: 100%; text-align: center;"> <tr> <td>1.67</td> <td>6.67</td> <td>3.33</td> <td>1.67</td> <td>ratio of mol ✓</td> </tr> </table> <p>to give <math>\text{CH}_4\text{N}_2\text{O}</math> ✓</p> | C          | H   | N              | O |  | 20.00/12.0 | 6.67/1.0 | 46.67/14.0 | 26.66/16.0 |  | 1.67 | 6.67 | 3.33 | 1.67 | ratio of mol ✓ | 2 | <p><b>ALLOW</b> 1.66 for C <b>OR</b> 1.66 for O</p> <p><b>IGNORE</b> Significant figures beyond the 3rd significant figure.<br/>(eg <b>ALLOW</b> 3.3335 for N <b>OR</b> 1.666 for C)</p> <p><b>ALLOW</b> ECF from incorrectly calculated ratio of mol.<br/><b>DO NOT ALLOW</b> ECF from using an atomic number <b>OR</b><br/>any original sums inverted (eg 12.00/20.00)<br/><b>ALLOW</b> any order of atoms</p> |
| C            | H  | N          | O   |                |   |  |            |          |            |            |  |      |      |      |      |                |   |  |
| 20.00/12.0   | 6.67/1.0   | 46.67/14.0 | 26.66/16.0  |                |   |  |            |          |            |            |  |      |      |      |      |                |   |  |
| 1.67         | 6.67   | 3.33       | 1.67  | ratio of mol ✓ |   |  |            |          |            |            |  |      |      |      |      |                |   |  |
| 2 (b)        | $\text{NH}_4^+$ ✓ $\text{NO}_3^-$ ✓  | 2          | Mark incorrect ions first   |                |   |  |            |          |            |            |  |      |      |      |      |                |   |  |
| 2 (c) (i)    | $\text{H}_3\text{PO}_4$ ✓  | 1          | <b>ALLOW</b> formula if seen as reactant in an equation<br><b>IGNORE</b> name |                |   |  |            |          |            |            |  |      |      |      |      |                |   |  |
| 2 (c) (ii)   | Calcium oxide <b>OR</b> calcium hydroxide <b>OR</b> calcium carbonate ✓  | 1          | <b>IGNORE</b> formulae<br><b>IGNORE</b> lime, quicklime and limestone         |                |   |  |            |          |            |            |  |      |      |      |      |                |   |  |
| <b>Total</b> |  | <b>6</b>   |   |                |   |  |            |          |            |            |  |      |      |      |      |                |   |  |

# June 2014 F321 Question Paper

**Joacim Rocklöv**



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