

Final study guide question and answers

What kind of equilibrium exists and under what conditions? **Chemical equilibrium** is a dynamic equilibrium between two or more states of matter. It is a state of balance between the forward and reverse reactions.

What is the difference between a reversible and an irreversible reaction? **Reversible reactions** are those that can proceed in both directions, while **irreversible reactions** proceed only in one direction.

What is the difference between a homogeneous and a heterogeneous equilibrium? **Homogeneous equilibrium** involves reactants and products in the same phase, while **heterogeneous equilibrium** involves reactants and products in different phases.

What is the difference between a closed and an open system? **Closed systems** allow energy to be exchanged with the surroundings, but not matter. **Open systems** allow both energy and matter to be exchanged with the surroundings.

What is the difference between a pure substance and a mixture? **Pure substances** have a fixed composition and distinct chemical and physical properties. **Mixtures** have a variable composition and do not have fixed chemical and physical properties.

What is the difference between a solid, a liquid, and a gas? **Solids** have a fixed shape and volume. **Liquids** have a fixed volume but no fixed shape. **Gases** have neither a fixed shape nor a fixed volume.

What is the difference between a crystalline and an amorphous solid? **Crystalline solids** have a regular, repeating arrangement of particles. **Amorphous solids** do not have a regular arrangement of particles.

What is the difference between a metal and a non-metal? **Metals** are good conductors of heat and electricity. **Non-metals** are poor conductors of heat and electricity.

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Nate Test Study Guide

Rachel S Tattersall



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