

Q1

The common oxidation states of Ti are

- (a) +2 and +3 (b) +3 and +4  
(c) -3 and -4 (d) +2, +3 and +4

Q2

When phenol is treated with excess bromine water, it gives

- (a) *m*-Bromophenol (b) *o*- and *p*-Bromophenols  
(c) 2,4-Dibromophenol (d) 2,4,6-Tribromophenol.

Q3

If 0.5 mol of  $\text{BaCl}_2$  is mixed with 0.2 mole of  $\text{Na}_3\text{PO}_4$ , find the maximum amount of  $\text{Ba}_3(\text{PO}_4)_2$  that can be formed.

- (a) 1 mole (b) 0.5 mole (c) 0.1 mole (d) 0.01 mole

Q4

Which is chemically most active non-metal?

- (a) S (b) O (c) F (d) N

Q5

The following species will not exhibit disproportionation reaction

- (a)  $\text{ClO}^-$  (b)  $\text{ClO}_2^-$  (c)  $\text{ClO}_3^-$  (d)  $\text{ClO}_4^-$

Q6

Which of the following has been used in the manufacture of non-inflammable photographic films?

- (a) Cellulose nitrate (b) Cellulose xanthate  
(c) Cellulose perchlorate (d) Cellulose acetate

Q7

Point out the wrong statement:

For a first order reaction

- (a) time for half-change ( $t_{1/2}$ ) is independent of initial concentration  
(b) change in the concentration unit does not change the rate constant ( $k$ )  
(c) time for half-change  $\times$  rate constant = 0.693  
(d) the unit of  $k$  is  $\text{mol}^{-1} \text{min}^{-1}$

Q8

Name of the structure of silicates in which three oxygen atoms of  $[\text{SiO}_4]^{4-}$  are shared.

- (a) Pyrosilicate  
(b) Sheet silicate  
(c) Linear chain silicate  
(d) Three dimensional silicate

Q9

In the following sequence of reactions



- (a) *n*-propyl amine (b) isopropylamine  
(c) ethylamine (d) ethylmethyl amine

Q10

Which of the following monosaccharide is a pentose?

- (a) Galactose (b) Glucose  
(c) Fructose (d) Arabinose

Q11

In Bohr's model, atomic radius of the first orbit is  $y$ , the radius of the 3<sup>rd</sup> orbit, is

- (a)  $y/3$  (b)  $y$   
(c)  $3y$  (d)  $9y$

Q12

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**L Reisser**



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## **Table of Contents Jee Sample Papers 2013**

1. Understanding the eBook Jee Sample Papers 2013
  - The Rise of Digital Reading Jee Sample Papers 2013
  - Advantages of eBooks Over Traditional Books
2. Identifying Jee Sample Papers 2013
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Jee Sample Papers 2013
  - User-Friendly Interface
4. Exploring eBook Recommendations from Jee Sample Papers 2013
  - Personalized Recommendations
  - Jee Sample Papers 2013 User Reviews and Ratings
  - Jee Sample Papers 2013 and Bestseller Lists
5. Accessing Jee Sample Papers 2013 Free and Paid eBooks
  - Jee Sample Papers 2013 Public Domain eBooks
  - Jee Sample Papers 2013 eBook Subscription Services
  - Jee Sample Papers 2013 Budget-Friendly Options
6. Navigating Jee Sample Papers 2013 eBook Formats

- ePub, PDF, MOBI, and More
  - Jee Sample Papers 2013 Compatibility with Devices
  - Jee Sample Papers 2013 Enhanced eBook Features
7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Jee Sample Papers 2013
    - Highlighting and Note-Taking Jee Sample Papers 2013
    - Interactive Elements Jee Sample Papers 2013
  8. Staying Engaged with Jee Sample Papers 2013
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Jee Sample Papers 2013
  9. Balancing eBooks and Physical Books Jee Sample Papers 2013
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Jee Sample Papers 2013
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Jee Sample Papers 2013
    - Setting Reading Goals Jee Sample Papers 2013
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Jee Sample Papers 2013
    - Fact-Checking eBook Content of Jee Sample Papers 2013
    - Distinguishing Credible Sources
  13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development
    - Exploring Educational eBooks
  14. Embracing eBook Trends
    - Integration of Multimedia Elements
    - Interactive and Gamified eBooks

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