

Interpreting Chemical Formulas

Instructions: Follow the example in the first row to record the number of atoms of each element in the compounds. After you have completed the table, answer the questions in the spaces provided.

The Chemical Formulas of Common Compounds

Name of Compound	Examples of Common Uses	Chemical Formula	Number of Atoms of Each Element
calcium carbonate	Chalk for writing on a chalkboard	CaCO_3	<ul style="list-style-type: none"> • 1 atom calcium (Ca) • 1 atom carbon (C) • 3 atoms oxygen (O)
sodium phosphate	Heavy-duty cleaner	Na_3PO_4	•
magnesium chloride	De-icer for roads	MgCl_2	•
monosodium glutamate (MSG)	Flavour-enhancer for foods	$\text{NaC}_5\text{H}_9\text{NO}_4$	•
hydrogen peroxide	Bleach and disinfectant	H_2O_2	•
glucose	Sweetener (glucose is a sugar)	$\text{C}_6\text{H}_{12}\text{O}_6$	•
carbon dioxide	Addition of fizz to pop	CO_2	•
freon-12	Refrigerator coolant (now banned because it harms the ozone layer)	CCl_2F_2	•
HFC-32 (difluoromethane)	Refrigerator coolant (now used because it does not harm the ozone layer)	CH_2F_2	•

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