

# Intro to Acids and Bases

Name: \_\_\_\_\_  
 Class: \_\_\_\_\_  
 Date: \_\_\_\_\_

## Table 1: Definitions of Acids and Bases

### Arrhenius Definitions

- acid substances that release hydrogen ions in aqueous solutions (H<sup>+</sup>) (acids H<sup>+</sup> gases formed in bases as the hydroxide ion, OH<sup>-</sup>, and H<sup>+</sup> (acids H<sub>2</sub>O) and conjugate bases)
- base substances which dissociate water molecules (OH<sup>-</sup>)

### Brønsted-Lowry Definitions

- acid substances that donate a proton (H<sup>+</sup>) from another
- base substances which accept a proton (H<sup>+</sup>) from another

## Table 2: Representative Acids and Bases (in water) with their respective reactions

Acids	Bases
1. $\text{HCl(aq)} + \text{H}_2\text{O(l)} \rightarrow \text{H}_3\text{O}^+(\text{aq}) + \text{Cl}^-(\text{aq})$	1. $\text{NaOH(aq)} + \text{H}_2\text{O(l)} \rightarrow \text{Na}^+(\text{aq}) + \text{OH}^-(\text{aq})$
2. $\text{HNO}_3 + \text{H}_2\text{O} \rightarrow \text{H}_3\text{O}^+ + \text{NO}_3^-$ (strong acid)	2. $\text{KOH} + \text{H}_2\text{O} \rightarrow \text{K}^+ + \text{OH}^-$
3. $\text{H}_2\text{SO}_4 + \text{H}_2\text{O} \rightarrow \text{H}_3\text{O}^+ + \text{HSO}_4^-$	3. $\text{NH}_4\text{OH} + \text{H}_2\text{O} \rightarrow \text{NH}_4^+ + \text{OH}^-$
4. $\text{H}_2\text{CO}_3 + \text{H}_2\text{O} \rightarrow \text{H}_3\text{O}^+ + \text{HCO}_3^-$	4. $\text{Ca(OH)}_2 + \text{H}_2\text{O} \rightarrow \text{Ca}^{2+} + 2\text{OH}^-$
5. $\text{H}_2\text{O} + \text{H}_2\text{O} \rightarrow \text{H}_3\text{O}^+ + \text{OH}^-$	5. $\text{LiOH} + \text{H}_2\text{O} \rightarrow \text{Li}^+ + \text{OH}^-$

## Critical Thinking Questions

1. Use your definitions for Table 1 that are appropriate with:

**(a) as Arrhenius acids and**

2. Use your definitions for Table 1 that are appropriate with:

**(b) as**

3. Use your definitions for Table 1 that are appropriate using your:

**(a) as**

4. Use your definitions for Table 1 that are appropriate using your:

**(a) as**

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8. Use your definitions for Table 1 that are appropriate using your:

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9. Use your definitions for Table 1 that are appropriate using your:

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**(a) as**

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**Clifford W. Hand, Harry Lyon Blewitt**

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