

## Comprehensive Examinations – 2023

(According to the syllabus of the Board of Secondary Education, Karachi.)

Time : 3 Hours

### CHEMISTRY (IX)

Max. Marks : 60

#### SECTION "A" (COMPULSORY) MULTIPLE CHOICE QUESTIONS (MCQs) (12 Marks)

Note: (i) Attempt all the questions.

(ii) Do not copy the questions. Write only the answer against the proper number of the question and its part.

(iii) Each question carries **ONE** mark.

Q.1 Choose the correct answer for each from the given options:

(i) Symbols of elements were suggested by:

- \* John Dalton                      \* J.J. Thomson                      \* J. J. Berzelius                      \* James Chadwick

(ii) This of the following can be separated by physical means:

- \* Element                      \* Compound                      \* Mixture                      \* none of these

(iii) This material of screen was used in Rutherford atomic model :

- \* Aluminium foil                      \* Zinc sulphide                      \* Sodium sulphide                      \* Aluminium sulphide

(iv)  $1s^2, 2s^2 2p^6, 3s^2 3p^3$  is the electronic configuration of:

- \*  $^{15}_7\text{N}$                       \*  $^{16}_8\text{O}$                       \*  $^{31}_{15}\text{P}$                       \*  $^{32}_{16}\text{S}$

(v) All Noble gases have complete octet except:

- \* He                      \* Ne                      \* Ar                      \* Kr

(vi) All transition elements are:

- \* Metals                      \* Non-metals                      \* Metalloids                      \* none of these

(vii) Atoms lose, gain or share electron(s) to achieve the electronic configuration of:

- \* Alkali metals                      \* Alkaline earth metals                      \* Halogens                      \* Noble gases

(viii) In NaCl crystal lattice, each Sodium ion is surrounded by:

- \* 6  $\text{Cl}^-$  ions                      \* 6  $\text{Na}^+$  ions                      \* 8  $\text{Cl}^-$  ions                      \* 8  $\text{Na}^+$  ions

(ix) It is the most energetic, super hot & super excited state of matter:

- \* BEC                      \* Liquid                      \* Gas                      \* Plasma

(x) Solution of known concentration is termed as:

- \* Aqueous solution                      \* Standard solution  
\* Dilute solution                      \* Concentrated solution

(xi) The hardness of diamond at Mohs scale is:

- \* 0                      \* 2.5                      \* 4.5                      \* 10

(xii) This colour is obtained from the compounds of Sodium:

- \* Crimson Red                      \* Golden Yellow                      \* Violet                      \* Lilac

# Model Paper For Practical Test

**N Colangelo**



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