

Chemistry – Unit 7 Review: Chemical Reactions

- ___ 1. Numbers placed in front of formulas in an equation are called
- ___ 2. New substances formed in a chemical reaction are called
- ___ 3. The total number of atoms represented by the formula $\text{Fe}_3(\text{PO}_4)_2$ is
- ___ 4. The symbol that indicates a substance dissolved in water is
- ___ 5. Two white powders are mixed in a flask. The bottom of the flask feels cooler. This an example of what kind of reaction?
- ___ 6. The charge on lead in PbSO_4 is



- _____ 7. How many more D atoms should be on the right side of the equation?
- _____ 8. What is the formula of the missing product molecule?
- _____ 9. What are particles on the left side of the arrow called?
10. Write the completed chemical equation for the reaction pictured.

Balance the following reactions.



Modeling Chemistry Answers

Christian Drosten



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