

# Grade 7

# Natural Sciences

## Term 4

### Exam paper 1

Section	Questions	Answers
<b>Section A</b> Name 3 elements: _____ <b>Instructions:</b> 1. Label each circle with an element name. 2. Write an atomic symbol under each name. 3. Write the mass number and atomic number of each element. 4. Write the number of protons and neutrons in each nucleus. 5. Write the number of electrons in each atom. 	<b>QUESTION 1</b> Diagram 1 shows a neutral atom with 11 protons and 11 electrons. The mass number of the atom is 23. 1.1 What is the best description for this atom? 1.2 What is the mass number of the atom? 1.3 What is the atomic number of the atom? 1.4 How many protons are there in the nucleus of the atom? 1.5 How many neutrons are there in the nucleus of the atom? 1.6 How many electrons are there in the atom? 1.7 How many shells does the atom have? 1.8 How many electrons are there in the outer shell of the atom? 1.9 How many protons are there in the nucleus of the atom? 1.10 How many neutrons are there in the nucleus of the atom? 1.11 How many electrons are there in the atom? 1.12 How many shells does the atom have? 1.13 How many electrons are there in the outer shell of the atom?	1.1 A neutral atom. 1.2 23 1.3 11 1.4 11 1.5 12 1.6 2 1.7 3 1.8 1 1.9 11 1.10 12 1.11 11 1.12 3 1.13 1



# Natural Science Grade 7 Question Papers

**Arturo Cuomo**



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