

Quiz & Worksheet - Strong Intermolecular Forces

1. Which of the following is true of intermolecular forces?

- The forces that exist between molecules.
- The forces that exist within a molecule.
- The force that bonds hydrogen with oxygen.
- The strongest force in chemistry.

2. Hydrogen bonds exist between X and Y. Which combination could be X and Y?

- X is water, H_2O
Y is ammonia, NH_3
- X is hydrogen, H_2
Y is water, H_2O
- X is oxygen, O_2
Y is hydrogen, H_2
- X is oxygen, O_2
Y is sulphur dioxide, SO_2

3. Identify the requirement for a dipole-dipole force to exist.

- The molecules need to be very close to each other.
- The molecules need to be very far apart from each other.
- The molecules need to include hydrogen.
- Dipole-dipole forces can happen in any molecule in any situation.

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