

**12. Molecular Geometry and Bonding Theories**

Week 11/12/2021 and Week 12/13/2021, Wednesday 12, 2021

How do we determine the shape of a molecule? To determine the geometry of a molecule, we use the following steps:

1. Determine the central atom.

1/12/2021

Make a table for Chapter 9 and 10 and use it as a reference for the next part of the course.

Molecular Geometry	
<p><b>Learning Objectives:</b> To determine the geometry of a molecule, we use the following steps:</p> <p>1. Determine the central atom.</p> <p>2. Determine the number of valence electrons around the central atom.</p> <p>3. Determine the number of lone pairs on the central atom.</p> <p>4. Determine the number of electron groups around the central atom.</p> <p>5. Determine the electron group geometry.</p> <p>6. Determine the molecular geometry.</p> <p>7. Determine the bond angles.</p>	<p style="text-align: center;">Molecular Geometry</p>
<p>1. Determine the central atom.</p>	<p>2. Determine the number of valence electrons around the central atom.</p>
<p>3. Determine the number of lone pairs on the central atom.</p>	
<p>4. Determine the number of electron groups around the central atom.</p>	
<p>5. Determine the electron group geometry.</p>	
<p>6. Determine the molecular geometry.</p>	
<p>7. Determine the bond angles.</p>	
<p>8. Determine the hybridization of the central atom.</p>	
<p>9. Determine the polarity of the molecule.</p>	
<p>10. Determine the intermolecular forces.</p>	

# Masteringchemistry Answer Key

**D Siedentop**



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