

Name: _____ () Class: ___ Date: _____

A. Multiple choice questions (10 marks)

1. Which of the following statements concerning vinegar is/are correct?

- (1) The process of forming hydrogen ions in vinegar is reversible.
(2) Neutralization occurs when sugar is added to vinegar.
(3) The pH of vinegar used in kitchen is around 1.

- A. (1) only
B. (2) only
C. (1) and (2) only
D. (2) and (3) only

2. 25.00 cm³ of 0.50 M lead(II) nitrate solution is mixed with 50.00 cm³ of 1.00 M sodium chloride solution. Insoluble lead(II) chloride is formed during mixing. What is the concentration of Cl⁻(aq) in the mixture?

- A. 0.33 M
B. 0.50 M
C. 0.75 M
D. 1.50 M

3. Consider the solutions W, X, Y, Z below:

W	100 cm ³ of 0.20 M HNO ₃ (aq)
X	50 cm ³ of 0.20 M HCl (aq)
Y	100 cm ³ of 0.20 M CH ₃ CO ₂ H (aq)
Z	50 cm ³ of 0.10 M NaOH (aq)

Which of the following statements is/are correct?

- A. The pH of Y equals to $-\log 0.2$.
B. Mixing W and Z gives a neutral solution.
C. The pH of the mixture of W and X is lower than that of W.
D. The pH of the mixture of W and X is lower than that of the mixture of X and Y.

4. Which of the following statements about ethanoic acid is correct?

- A. It is a strong acid.
B. Its empirical formula is CH₂O.
C. It forms a salt NaCH₂COO when reacting with NaOH (aq).

Modern Chemistry Ch14 Review Answers

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