

Chapter Test A

Chapter: Chemical Equations and Reactions

Do the work carefully, using the letter of the letter of choice that best completes each statement or best answers each question.

1. Which statement is **not** correct?
 - a. The number of atoms is conserved.
 - b. The number of molecules is not conserved.
 - c. The number of molecules is not conserved, but the number of atoms is conserved.
 - d. None of the above; all three statements are correct.
2. In the reaction described by a chemical equation,

$$\text{reactants} \rightarrow \text{products} + \text{products} + \text{products}$$
 - a. all the reactants are products.
 - b. all the reactants are products.
 - c. all the reactants are products.
 - d. None is correct.
3. Which of the following relationships is **not** correct?
 - a. The products of a reaction are the substances with matter (as bonds) after the reaction.
 - b. All of the reactants are products.
 - c. The reactants of a reaction are the substances before the reaction begins.
 - d. The reactants of a reaction are the substances with matter (as bonds) before the reaction.
4. Which is **not** a correct statement?
 - a. A chemical equation is a shorthand way of writing a chemical reaction.
 - b. A chemical equation is a shorthand way of writing a chemical reaction.
 - c. A chemical equation is a shorthand way of writing a chemical reaction.
 - d. All of the above.
5. In a chemical equation,
 - a. reactants are products.
 - b. reactants are products.
 - c. reactants are products.
 - d. reactants are products.

Modern Chemistry Test B Answer Key Chapter20

William C. Doster



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