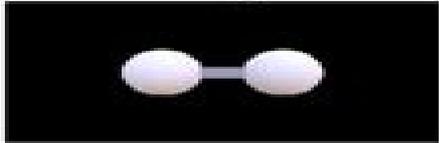
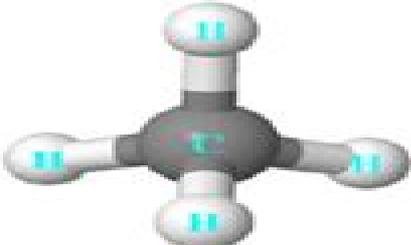


## Molecule Building Simulation

- 1) Find the website for PHET simulations.
- 2) Open the program entitled "Molecule Shapes" under Chemistry Programs.
- 3) Build the Lewis structure of the following molecules before building them on the computer.
- 4) Build the molecules on the simulation (make sure to include lone pairs of electrons on the central atom)
- 5) Check the boxes for "Molecule Geometry" and "Show bond angles" and fill in the following chart. Give a brief sketch of what the model looks like in the final box.

Molecule	Lewis Drawing	Bond <sup>o</sup>	Shape Name	3-D Drawing/Picture
H <sub>2</sub>	H:H	180	linear	
CH <sub>4</sub>	<pre>       H         H : C : H               H           </pre>	109.5	tetrahedral	
NH <sub>3</sub>	<pre>       H   N   H                       H   H   H           </pre>	107.8	trigonal pyramidal	
H <sub>2</sub> O	<pre>       O         \       H  H           </pre>	104.5	bent	

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**Robert R. Redfield**



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