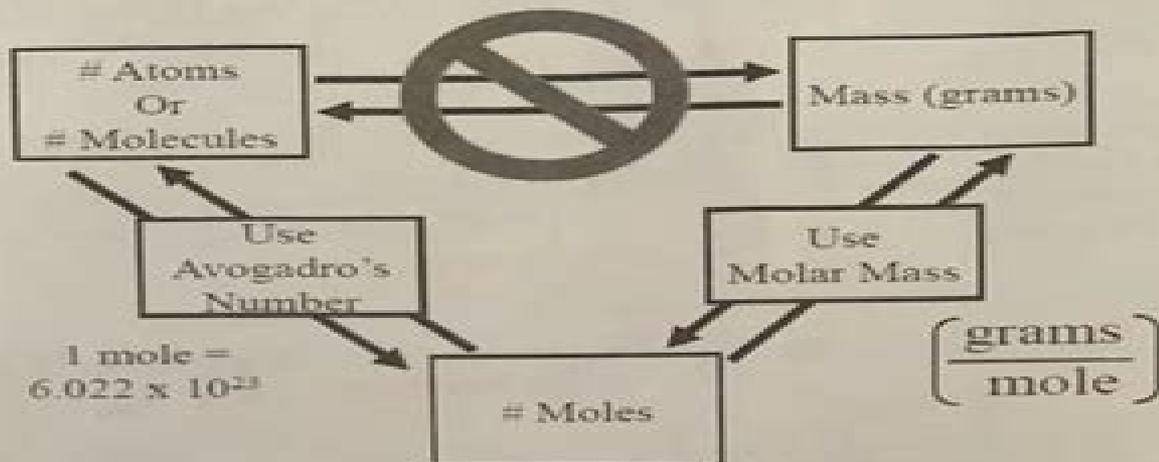


Moles, Molecules, and Grams Worksheet and Key



1) How many moles are there in 24.0 grams of FeF_3 ?

Fe	1 X 55.85	55.85 g/mol	24.0 grams FeF_3	1 mol FeF_3
F	3 X 19.00	+ 57.00 g/mol		
		112.85 g/mol	360	112.85 grams FeF_3
			= 0.212671688	0.213 mole FeF_3

2) How many moles are there in 458 grams of Na_2SO_4 ?

Na	2 X 22.99	45.98 g/mol	458 grams Na_2SO_4	1 mol Na_2SO_4
S	1 X 32.07	+ 32.07 g/mol		
O	4 X 16.00	+ 64.00 g/mol	360	142.05 grams Na_2SO_4
		142.05 g/mol	= 3.224216825	3.22 mole Na_2SO_4

3) How many grams are there in 2.30×10^{24} atoms of silver?

Ag	1 X 107.87	107.87 g/mol	360	1 mol Ag
	2.30×10^{24} atoms Ag			
		6.022×10^{23} atom Ag	107.87 grams Ag	1 mol Ag
			= 411.9910329	412 grams Ag

4) How many grams are there in 7.40 moles of AgNO_3 ?

Ag	1 X 107.87	107.87 g/mol	7.40 moles AgNO_3	169.88 grams AgNO_3
N	1 X 14.01	+ 14.01 g/mol		
O	3 X 16.00	+ 48.00 g/mol	360	1 moles AgNO_3
		171.88 g/mol	= 1252.112	1260 grams AgNO_3

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Raffaela Di Napoli



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