

## Assessment 3

**Chapter Test B****Chapter: Chemical Equations and Reactions**

**Part 1** In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- \_\_\_\_\_ 1. The production of a slightly soluble solid compound by a double-displacement reaction is called the formation of a
- gas.
  - precipitate.
  - redox reaction.
  - halogen.
- \_\_\_\_\_ 2. To balance a chemical equation, it is permissible to adjust the
- coefficients.
  - subscripts.
  - formula of the products.
  - number of products.
- \_\_\_\_\_ 3. In a chemical equation, the symbol (aq) indicates that the substance is
- solid.
  - dissolved in water.
  - an acid.
  - aqueous.
- \_\_\_\_\_ 4. The tendency for a replacement reaction to occur depends on the
- activity between any two elements in the activity series themselves.
  - temperature difference.
  - valence electrons and mass of.
  - activity between any two elements in the activity series themselves.
- \_\_\_\_\_ 5. The coefficients in a chemical equation
- indicate the number of moles of each substance that react.
  - show the number of grams of each substance that react.
  - are the molar masses of the substances.
  - show the valence electrons for each atom.
- \_\_\_\_\_ 6. If metal X is more reactive than metal Y in the activity series, then
- X will replace ions of Y in solution.
  - Y will replace ions of X in solution.
  - Y will form oxides only indirectly.
  - X will react with cold water.

# Holt Chemistry Answer Key Chap 8

**Edward Florian Neuzil**



## Holt Chemistry Answer Key Chap 8:

**Holt Chemistry** Ralph Thomas Myers,2004      **Holt Chemistry File** ,1998 This reference is a must for students who need extra help reteaching or extra practice The guide moves students through the same concepts as the text but at a slower pace More descriptive detail along with visual algorithms provides a more structured approach Each chapter closes with a large bank of practice problems Book jacket      **Books in Print Supplement** ,2002      Children's Books in Print, 2007 ,2006      *Instructor's Guide and Answer Keys to Accompany The Holt Handbook* ,2002      **Journal** Royal Society for the Encouragement of Arts, Manufactures and Commerce,1868      *Introduction to Modern Chemistry* Edward Florian Neuzil,1968      Modern Chemistry Raymond E. Davis,1999 2000 2005 State Textbook Adoption Rowan Salisbury      Essentials of Modern Chemistry Charles Elwood Dull,1918      *JOURNAL OF THE SOCIETY OF ARTS AND OF THE INSTITUTIONS IN UNION* ,1868      **Journal of the Royal Society of Arts** Royal Society for the Encouragement of Arts, Manufactures and Commerce,1868      *Journal of the Society of Arts* Royal Society of Arts (Great Britain),1868      **Journal of the Royal Society of Arts** Royal Society of Arts (Great Britain),1868      Forthcoming Books Rose Army,2003-04      *Principles of Modern Chemistry* David W. Oxtoby, Norman H. Nachtrieb,1996 PRINCIPLES OF MODERN CHEMISTRY has dominated the honors and high mainstream general chemistry courses and is considered the standard for the course The fifth edition is a substantial revision that maintains the rigor of previous editions but reflects the exciting modern developments taking place in chemistry today Authors David W Oxtoby and H P Gillis provide a unique approach to learning chemical principles that emphasizes the total scientific process from observation to application placing general chemistry into a complete perspective for serious minded science and engineering students Chemical principles are illustrated by the use of modern materials comparable to equipment found in the scientific industry Students are therefore exposed to chemistry and its applications beyond the classroom This text is perfect for those instructors who are looking for a more advanced general chemistry textbook      **Journal** ,1868      **Journal of the Society of Arts** ,1868      *Lab Experiments Modern Chemistry* Harold Clark Metcalfe,1982      **Study and Problem Solving Guide to Accompany Principles of Modern Chemistry, Oxtoby/Nachtrieb** Wade A. Freeman,1986      **Modern Chemistry** H. Clark Metcalfe, Joseph F. Castka,1966

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