

CHAPTER 7 TEST FOR ORGANIC CHEMISTRY QUESTIONS WITH COMPLETE SOLUTIONS

an example of chemical changes is _____

- A. water changing to ice
- B. ice melting to water
- C. the oxidation of haircolor products
- D. temporary haircolor - Answer-C. the oxidation of haircolor products

_____ refers to either the loss of oxygen or the addition of hydrogen.

- A. application
- B. reduction
- C. reaction
- D. oxidation - Answer-B. reduction

when using a permanent wave neutralizer _____

- A. H₂O₂ loses oxygen
- B. the waving solution is oxidized
- C. haircolor gains oxygen from H₂O₂
- D. hair is oxidized by removing hydrogen - Answer-D. hair is oxidized by removing hydrogen

when oxygen is combined with a substances, the substances is _____

- A. oxidized
- B. reduced
- C. United
- D. blended - Answer-A. oxidized

oxidation and reduction always occur simultaneously and are referred to as a _____ reaction

- A. rapid oxidation
- B. endothermic
- C. redox
- D. endothermic - Answer-C. redox

what type of chemical reaction requires the absorption of energy or heat from an external source for the reaction to actually occur?

- A. endothermic
- B. acid-alkali neutralization
- C. exothermic
- D. redox - Answer-A. endothermic

which of the following is an example of a pure substances?

- A. concrete
- B. Aluminum foil
- C. powder

Holt Chemistry Chapter 7 Test

Michal Rosen-Zvi



Holt Chemistry Chapter 7 Test:

Holt Chemistry Ralph Thomas Myers,2004 **Books in Print Supplement** ,2002 Holt Chemistry File ,1998 This reference is a must for students who need extra help reteaching or extra practice The guide moves students through the same concepts as the text but at a slower pace More descriptive detail along with visual algorithms provides a more structured approach Each chapter closes with a large bank of practice problems Book jacket *Children's Books in Print, 2007* ,2006 Holt Biology Holt Rinehart & Winston,2003-08 **Children's Books in Print** R R Bowker Publishing,Bowker,1999-12 El-Hi Textbooks and Serials in Print ,1985 Holt Algebra 1 2003 Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2003 El-Hi Textbooks and Serials in Print, 1996 R R Bowker Publishing,Bowker,1996-02 Trace Environmental Quantitative Analysis Paul R. Loconto,2020-12-27 A thorough and timely update this new edition presents principles techniques and applications in this sub discipline of analytical chemistry for quantifying traces of potentially toxic organic and inorganic chemical substances found in air soil fish and water as well as serum plasma urine and other body fluids The author addresses regulatory aspects calibration verification and the statistical treatment of analytical data including instrument detection limits quality assurance quality control sampling and sample preparation and techniques that are used to quantify trace concentrations of organic and inorganic chemical substances Key Features Fundamental principles are introduced for the more significant experimental approaches to sample preparation Principles of instrumental analysis determinative techniques for trace organics and trace inorganics analysis An introduction to the statistical treatment of trace analytical data How to calculate instrument detection limits based on weighted least squares confidence band calibration statistics Includes an updated series of student tested experiments

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