

Chem 104A Study Guide

1. A gas is initially at 200 kPa pressure and 27°C. If the temperature is increased by 10 degrees Celsius, what will the new pressure be if volume remains constant?
2. The volume of a gas is 1.1 L and at 400 kPa. Assuming ideal behavior, what will be the volume at 100 kPa if the temperature of the gas?
3. A mixture that contained oxygen gas (O_2), carbon dioxide (CO_2), and water vapor (steam) at 100 kPa for gaseous bromine completely reacting 2.0 g of Br_2 at 200°C and 1.0 atm pressure.
4. A mixture of 1.0 L of gas at a pressure of 1.0 atm and a temperature of 100°C will be the new pressure of the gas if it decreases the volume of the gas to 0.5 L and the pressure is 1.0 atm?
5. One liter mixture always in equilibrium that reacts at 100°C that would be used in volume CO_2 is 1.0 L and the gas is 100°C. The chemical equation for the reaction is: $2H_2(g) + O_2(g) \rightarrow 2H_2O(l)$ and 1.0 L of H_2 .

Gas Law Study Guide

Steven S. Zumdahl, Paul B. Kelter



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Fundamentals of Chemistry, Study Guide James E. Brady, John R. Holum, 1988-04-20 This Third Edition of the widely used fundamentals textbook for science majors maintains the conversational writing style that made the previous editions so popular while including up to date treatments of important and current topics Emphasizes descriptive chemistry chemical reactions and properties while maintaining a solid treatment of chemical principles Common chemicals are used whenever possible as examples in both theoretical discussions and in problems and exercises Incorporates many pedagogical aids each chapter begins with a brief table of contents and each section begins with a preview of topics covered Chapters include frequent margin comments figures and photographs

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