

Gas laws practice test

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- _____ 1. Pressure is the force per unit
a. volume. c. length.
b. surface area. d. depth.
- _____ 2. Why does a can collapse when a vacuum pump removes air from the can?
a. The inside and outside forces balance out and crush the can.
b. The unbalanced outside force from atmospheric pressure crushes the can.
c. The atmosphere exerts pressure on the inside of the can and crushes it.
d. The vacuum pump creates a force that crushes the can.
- _____ 3. If the height of mercury in a barometer at 0°C is less than 760 mm Hg, then
a. the atmospheric pressure is less than standard atmospheric pressure.
b. the atmospheric pressure is greater than standard atmospheric pressure.
c. the atmospheric pressure is equal to standard atmospheric pressure.
d. the atmospheric pressure cannot be determined.
- _____ 4. Convert the pressure 0.840 atm to mm Hg.
a. 365 mm Hg c. 638 mm Hg
b. 437 mm Hg d. 780 mm Hg
- _____ 5. Convert the pressure 1.30 atm to kPa.
a. 2 kPa c. 132 kPa
b. 115 kPa d. 245 kPa
- _____ 6. Standard temperature is exactly
a. 100°C. c. 0°C.
b. 273°C. d. 0 K.
- _____ 7. Three samples of gas each exert 740. mm Hg in separate 2 L containers. What pressure do they exert if they are all placed in a single 2 L container?
a. 247 mm Hg c. 1.48×10^3 mm Hg
b. 740 mm Hg d. 2.22×10^3 mm Hg
- _____ 8. If the temperature remains constant, V and P represent the original volume and pressure, and V' and P' represent the new volume and pressure, what is the mathematical expression for Boyle's law?
a. $P'V = V'P$ c. $VP = V'P'$
b. $VV' = PP'$ d. $V' = \frac{VP}{P'}$
- _____ 9. A sample of oxygen occupies 560. mL when the pressure is 800.00 mm Hg. At constant temperature, what volume does the gas occupy when the pressure decreases to 700.0 mm Hg?
a. 80.0 mL c. 600. mL
b. 490. mL d. 640. mL
- _____ 10. If the temperature of a fixed quantity of gas decreases and the pressure remains unchanged,
a. its volume increases. c. its volume decreases.
b. its volume is unchanged. d. its density decreases.

Gas Law Test Multiple Choice Answers

John Bird, Carl Ross



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Gas Law Test Multiple Choice Answers Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Gas Law Test Multiple Choice Answers**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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Table of Contents Gas Law Test Multiple Choice Answers

1. Understanding the eBook Gas Law Test Multiple Choice Answers
 - The Rise of Digital Reading Gas Law Test Multiple Choice Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Gas Law Test Multiple Choice Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gas Law Test Multiple Choice Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gas Law Test Multiple Choice Answers
 - Personalized Recommendations
 - Gas Law Test Multiple Choice Answers User Reviews and Ratings
 - Gas Law Test Multiple Choice Answers and Bestseller Lists
5. Accessing Gas Law Test Multiple Choice Answers Free and Paid eBooks

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 - Gas Law Test Multiple Choice Answers Budget-Friendly Options
6. Navigating Gas Law Test Multiple Choice Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Gas Law Test Multiple Choice Answers Compatibility with Devices
 - Gas Law Test Multiple Choice Answers Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gas Law Test Multiple Choice Answers
 - Highlighting and Note-Taking Gas Law Test Multiple Choice Answers
 - Interactive Elements Gas Law Test Multiple Choice Answers
 8. Staying Engaged with Gas Law Test Multiple Choice Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gas Law Test Multiple Choice Answers
 9. Balancing eBooks and Physical Books Gas Law Test Multiple Choice Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gas Law Test Multiple Choice Answers
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Gas Law Test Multiple Choice Answers
 - Setting Reading Goals Gas Law Test Multiple Choice Answers
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Gas Law Test Multiple Choice Answers
 - Fact-Checking eBook Content of Gas Law Test Multiple Choice Answers
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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