

13. Consider the following statements
- Atmospheric pressure is less at the top of a mountain than at sea level
 - Gases are much more compressible than solids or liquids
 - When the atmospheric pressure increases the height of the mercury column rises
- Select the correct statement
- I and II
 - II and III
 - I and III
 - I, II and III
14. Compressibility factor for CO_2 at 400 K and 71.0 bar is 0.8697. The molar volume of CO_2 under these conditions is
- 22.04 dm³
 - 2.24 dm³
 - 0.41 dm³
 - 19.5 dm³
15. If temperature and volume of an ideal gas is increased to twice its values, the initial pressure P becomes
- 4P
 - 2P
 - P
 - 3P
16. At identical temperature and pressure, the rate of diffusion of hydrogen gas is $3\sqrt{3}$ times that of a hydrocarbon having molecular formula $\text{C}_n\text{H}_{2n-2}$. What is the value of n ?
- 8
 - 4
 - 3
 - 1
17. Equal moles of hydrogen and oxygen gases are placed in a container, with a pin-hole through which both can escape what fraction of oxygen escapes in the time required for one-half of the hydrogen to escape. (NEET phase I)
- $\frac{3}{8}$
 - $\frac{1}{2}$
 - $\frac{1}{8}$
 - $\frac{1}{4}$
18. The variation of volume V , with temperature T , keeping pressure constant is called the coefficient of thermal expansion i.e. $\alpha = \frac{1}{V} \left(\frac{\partial V}{\partial T} \right)_P$. For an ideal gas α is equal to
- T
 - $\frac{1}{T}$
 - P
 - none of these
19. Four gases P, Q, R and S have almost same values of 'b' but their 'a' values (a, b are Vander Waals Constants) are in the order $Q < R < S < P$. At a particular temperature, among the four gases the most easily liquefiable one is
- P
 - Q
 - R
 - S
20. Maximum deviation from ideal gas is expected from (NEET)
- $\text{CH}_4(\text{g})$
 - $\text{NH}_3(\text{g})$
 - $\text{H}_2(\text{g})$
 - $\text{N}_2(\text{g})$
21. The units of Vander Waals constants 'b' and 'a' respectively
- mol L⁻¹ and L atm² mol⁻²
 - mol L and L atm mol²
 - mol⁻¹L and L² atm mol⁻²
 - none of these

Gaseous State Multiple Choice Questions

AN Whitehead



Gaseous State Multiple Choice Questions:

Reviewing **Gaseous State Multiple Choice Questions**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Gaseous State Multiple Choice Questions**," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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