

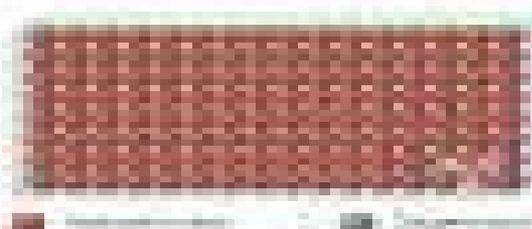


Student Exploration: Half-life

Essential Question: How does the half-life of a substance change as the amount of substance decreases?

1. Use the simulation to explore the relationship between half-life and amount of substance. The simulation starts with 100% of substance. **1000000** atoms. Record any patterns you see. **1000000** atoms. Record any patterns you see. **1000000** atoms. Record any patterns you see.

2. The half-life of a substance is the time it takes for half of the substance to decay.



<p>Modeling: Mathematical:</p>	<p>Use the simulation:</p> <ul style="list-style-type: none"> • The simulation starts with 100% of substance. Record any patterns you see. • Record the half-life of the substance and the number of atoms left.
--	---



Question: How does the half-life of a substance change as the amount of substance decreases?

- 1. Calculate** the half-life of the substance. Use the simulation to find the half-life of the substance. **1000000** atoms. Record any patterns you see. **1000000** atoms. Record any patterns you see. **1000000** atoms. Record any patterns you see.
- 2. Use** the simulation to find the half-life of the substance. Record any patterns you see. **1000000** atoms. Record any patterns you see. **1000000** atoms. Record any patterns you see. **1000000** atoms. Record any patterns you see.
- 3. Describe** the half-life of the substance. Use the simulation to find the half-life of the substance. Record any patterns you see. **1000000** atoms. Record any patterns you see. **1000000** atoms. Record any patterns you see. **1000000** atoms. Record any patterns you see.



Graph showing the decay of a substance over time.



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 0.444 mol of CoCl_2 in 0.654 l of solution
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 b 4.25 g of NH_3 in 0.500 l of solution the concentration of NH_3 in household ammonia

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$$\frac{\text{mass of a component}}{\text{mass of component in solution total mass of solution}} \times 100$$
 end array

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