

**MIXED Stoichiometry Practice Review Problems**

STEP 1: The following reaction will be used for this entire worksheet, please complete the balancing before going on.



STEP 2: Calculator the molar mass of each compound in the balanced reaction and write the values above the balanced reaction.

STEP 3: For aluminum sulfate, find the % of the metal in the compound. Please be sure to show your work. (Hint: W5.1 work)

**Al is the only metal, S and O are nonmetals. There are no metalloids in the formula.**

$$\% \text{ Al} = \frac{\text{total mass Al}}{\text{molar mass Al}_2(\text{SO}_4)_3} \times 100 \% = \frac{53.96 \text{ g}}{342.17 \text{ g}} \times 100 \% = 15.77 \%$$

STEP 4: Complete the following problems using **a factor chart with units & chemicals**. These problems could require ONE ratio, TWO ratios or even THREE ratios. Look for clues & use your mole map if necessary.

**PROBLEMS**

1) How many **grams of calcium sulfate** are produced from reacting **2.59 moles of aluminum sulfate**?

$$2.59 \text{ mol Al}_2(\text{SO}_4)_3 \times \frac{3 \text{ mol CaSO}_4}{1 \text{ mol Al}_2(\text{SO}_4)_3} \times \frac{136.15 \text{ g CaSO}_4}{1 \text{ mol CaSO}_4} = 1057.89 \text{ g CaSO}_4$$

moles A  $\longrightarrow$  moles B  $\longrightarrow$  grams B

# Mixed Problems Stoichiometry With Answers

**James E. Brady, Fred Senese**



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by a chemist who has some expertise in the specific technique discussed has done some research on the technique and has applied the technique in a chemistry course

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**Chemistry** James E. Brady,Fred Senese,2004-02-04 Publisher Description

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**Study Guide, Introductory Chemistry, a Foundation ; Introductory Chemistry ; Basic Chemistry** Iris Stovall,Donald J. DeCoste,2000

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In a world eaten by screens and the ceaseless chatter of instantaneous transmission, the melodic splendor and mental symphony developed by the prepared word often diminish into the back ground, eclipsed by the constant sound and disturbances that permeate our lives. However, set within the pages of **Mixed Problems Stoichiometry With Answers** a charming fictional value overflowing with raw feelings, lies an immersive symphony waiting to be embraced. Crafted by an outstanding musician of language, that charming masterpiece conducts viewers on a psychological trip, well unraveling the hidden melodies and profound impact resonating within each carefully constructed phrase. Within the depths of the moving analysis, we shall explore the book is main harmonies, analyze their enthralling writing fashion, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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