

**The Earth's early atmosphere**

- 1.1 Recall that the gases produced by volcanic activity formed the Earth's early atmosphere
- 1.2 Recall that the early atmosphere contained a little or no oxygen & a large amount of carbon dioxide & water vapour and small amounts of other gases
- 1.3 Explain why there are different sources of information about the development of the atmosphere which makes it difficult to be precise about the evolution of the atmosphere
- 1.4 Describe how condensation of water vapour formed oceans
- 1.5 Describe how the amount of carbon dioxide in the atmosphere was reduced by:
  - a) the distribution of carbon dioxide into the oceans & the later incorporation of the dissolved carbon dioxide into marine organisms which eventually formed calcareous rocks
  - b) Explain how the growth of primitive plants used carbon dioxide and released oxygen by photosynthesis and consequently the amount of oxygen in the atmosphere gradually increased
- 1.7 Investigate the proportion of oxygen in the atmosphere
- 1.8 Describe the current composition of the atmosphere and interpret data sources showing this information
- 1.9 Demonstrate an understanding of how small changes in the atmosphere occur through a volcanic activity & human activity including the burning of fossil fuels, farming and deforestation

**1. The composition of the early atmosphere**



Figure 1: The composition of the early atmosphere

Figure 2: The composition of the early atmosphere

Year	Oxygen (%)	Nitrogen (%)	Carbon Dioxide (%)
1800	20.9	78.1	0.1
1900	20.9	78.1	0.1
2000	20.9	78.1	0.1
2010	20.9	78.1	0.1



- 2. Complete the sentence by adding a word (B) in the box next to your answer
- a) The percentage of carbon dioxide in the Earth's atmosphere today is \_\_\_\_\_.

- A greater than 10%
- B 10%
- C 20%
- D less than 10%

b) The percentage of carbon dioxide in the Earth's atmosphere today is less than that in the Earth's earliest atmosphere. Explain what has caused the percentage of carbon dioxide to decrease.

\_\_\_\_\_

\_\_\_\_\_

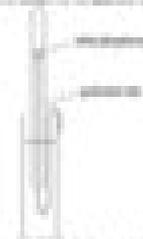
c) Carbon dioxide and other gases in the atmosphere help to keep the Earth warm. What are these gases called for Earth warm?

\_\_\_\_\_

\_\_\_\_\_



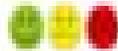
3. The diagram shows a gas cylinder with a piston. The temperature of the gas inside the cylinder is 300 K. The piston is pushed down and the temperature of the gas inside the cylinder is 400 K. The volume of the gas inside the cylinder is 1.0 m<sup>3</sup>. The pressure of the gas inside the cylinder is 1.0 × 10<sup>5</sup> Pa. The mass of the gas inside the cylinder is 1.0 kg. The density of the gas inside the cylinder is 1.0 kg m<sup>-3</sup>. The specific heat capacity of the gas is 1.0 J kg<sup>-1</sup> K<sup>-1</sup>. The work done on the gas is 1.0 J. The heat energy transferred to the gas is 1.0 J. The change in internal energy of the gas is 1.0 J.



The work done on the gas is 1.0 J. The heat energy transferred to the gas is 1.0 J. The change in internal energy of the gas is 1.0 J.

\_\_\_\_\_

\_\_\_\_\_



4. a) What is the composition of the early atmosphere?

- A 10% oxygen, 10% nitrogen, 80% carbon dioxide
- B 20% oxygen, 78% nitrogen, 0.1% carbon dioxide
- C 10% oxygen, 10% nitrogen, 80% carbon dioxide
- D 20% oxygen, 78% nitrogen, 0.1% carbon dioxide

b) The composition of carbon dioxide in the Earth's atmosphere today is 0.1%. Explain what has caused the percentage of carbon dioxide to decrease.

c) Explain how carbon dioxide is removed from the atmosphere.

\_\_\_\_\_

\_\_\_\_\_

d) Explain how carbon dioxide is added to the atmosphere.

\_\_\_\_\_

\_\_\_\_\_

e) How would the Earth have got its early atmosphere?

\_\_\_\_\_

\_\_\_\_\_



**CI Higher Topic Questions**

**Question 1 (10 marks)**

The diagram shows a gas cylinder with a piston. The temperature of the gas inside the cylinder is 300 K. The piston is pushed down and the temperature of the gas inside the cylinder is 400 K. The volume of the gas inside the cylinder is 1.0 m<sup>3</sup>. The pressure of the gas inside the cylinder is 1.0 × 10<sup>5</sup> Pa. The mass of the gas inside the cylinder is 1.0 kg. The density of the gas inside the cylinder is 1.0 kg m<sup>-3</sup>. The specific heat capacity of the gas is 1.0 J kg<sup>-1</sup> K<sup>-1</sup>. The work done on the gas is 1.0 J. The heat energy transferred to the gas is 1.0 J. The change in internal energy of the gas is 1.0 J.

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