

F325 Acids, Bases and Buffers test ms

1. (a) completely dissociates/ionised (1) 2  
 proton donor (1)
- (b)  $\text{NO}_3^-$  (1) 1
- (c) (i)  $\text{pH} = -\log[\text{H}^+] / -\log(0.015)$  (1) = 1.82 / 1.8 (1) (Not 2) 2  
 (ii)  $[\text{H}^+] = 0.0075 \text{ mol dm}^{-3}$   
 $\text{pH} = -\log(0.0075) = 2.12 / 2.1$  (1) 1

[6]

2. (a) (i)  $\text{H}^+$ /proton donor (1) 1  
 (ii) partially dissociates/ionises (1) 1
- (b)  $\text{C}_6\text{H}_5\text{OH}(\text{aq}) + \text{OH}^-(\text{aq}) \rightleftharpoons \text{C}_6\text{H}_5\text{O}^-(\text{aq}) + \text{H}_2\text{O}(\text{l})$   
 acid 1 base 2 (1) base 1 acid 2 (1)  
 1 mark for each acid-base pair
- (c) (i)  $K_a = \frac{[\text{C}_6\text{H}_5\text{O}^-][\text{H}^+]}{[\text{C}_6\text{H}_5\text{OH}]}$  (1) 1  
 (ii) concentration =  $38/94$  (1) =  $0.40 \text{ mol dm}^{-3}$  (1)  
 (first mark for  $M_r$  of phenol – incorrect answer here will  
 give ecf for remainder of question)

$$1.3 \times 10^{-10} = \frac{[\text{H}^+(\text{aq})]^2}{0.40} \quad (1)$$

( '=' sign is acceptable)

$$[\text{H}^+] = \sqrt{(1.3 \times 10^{-10}) \times (0.40)} = 7.2 \times 10^{-6} \text{ mol dm}^{-3} \quad (1)$$

$$\text{pH} = -\log[\text{H}^+] = -\log 7.2 \times 10^{-6} = 5.14 \quad (1) \quad 5$$

3 marks:  $[\text{H}^+]$  (1), pH expression (1), calc of pH from  $[\text{H}^+]$  (1)

**Common errors:**

Without square root, answer = 10.28 (1)(1)(0)

Use of 38 as molar concentration does not score 1<sup>st</sup> 2 marks.

This gives an answer of 4.15 for 3 marks (1)(1)(1)

- (d)  $[\text{H}^+(\text{aq})] = 1.99 \times 10^{-6} \text{ mol dm}^{-3}$  (1)  
 $[\text{C}_6\text{H}_5\text{O}^-(\text{aq})] = K_a [\text{C}_6\text{H}_5\text{OH}(\text{aq})] / [\text{H}^+(\text{aq})]$  (1)  
 $[\text{C}_6\text{H}_5\text{O}^-(\text{aq})] = 0.013 \text{ mol dm}^{-3}$  (1)

*Calculation should use half the original concentration of phenol to find the concentration of sodium phenoxide in the buffer. This should then be doubled back up again.*

*Do not penalise an approach that uses the original concentration of phenol in the expression above.*

[13]

# F325 Synoptic Questios

**JS Bruner**



**F325 Synoptic Questios:**

## Unveiling the Magic of Words: A Report on "F325 Synoptic Questios"

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### Table of Contents F325 Synoptic Questios

1. Understanding the eBook F325 Synoptic Questios
  - The Rise of Digital Reading F325 Synoptic Questios
  - Advantages of eBooks Over Traditional Books
2. Identifying F325 Synoptic Questios
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an F325 Synoptic Questios
  - User-Friendly Interface
4. Exploring eBook Recommendations from F325 Synoptic Questios
  - Personalized Recommendations
  - F325 Synoptic Questios User Reviews and Ratings
  - F325 Synoptic Questios and Bestseller Lists

5. Accessing F325 Synoptic Questions Free and Paid eBooks
  - F325 Synoptic Questions Public Domain eBooks
  - F325 Synoptic Questions eBook Subscription Services
  - F325 Synoptic Questions Budget-Friendly Options
6. Navigating F325 Synoptic Questions eBook Formats
  - ePub, PDF, MOBI, and More
  - F325 Synoptic Questions Compatibility with Devices
  - F325 Synoptic Questions Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of F325 Synoptic Questions
  - Highlighting and Note-Taking F325 Synoptic Questions
  - Interactive Elements F325 Synoptic Questions
8. Staying Engaged with F325 Synoptic Questions
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers F325 Synoptic Questions
9. Balancing eBooks and Physical Books F325 Synoptic Questions
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection F325 Synoptic Questions
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine F325 Synoptic Questions
  - Setting Reading Goals F325 Synoptic Questions
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of F325 Synoptic Questions
  - Fact-Checking eBook Content of F325 Synoptic Questions
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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