

Name: Date:

Student Exploration: Limiting Reactants

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: chemical equation, chemical formula, chemical reaction, coefficient, limiting reactant, molecule, product, reactant, subscript

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

Imagine you and your friends are making hot dogs. A complete hot dog consists of a wiener and a bun. At the store, you buy four packages of eight wieners and three bags of 10 buns.

1. How many total hot dogs can you make?

2. Which ingredient limited the number of hot dogs you could make?

3. Which ingredient will you have leftovers of?

Gizmo Warm-up

Just as ingredients can be put together to make a new food, substances can combine during a **chemical reaction** to produce new substances. The substances that undergo change are called **reactants**. The new substances are **products**.

Sometimes during a chemical reaction, one type of reactant will be used up before the other reactants. This reactant is the **limiting reactant**. Using the *Limiting Reactants* Gizmo, you can determine which reactant is limiting in various scenarios.

To begin, make sure **H₂ + O₂ becomes H₂O** is selected. The small "2" in H₂, O₂, and H₂O is a **subscript**. Subscripts represent the number of atoms in a **molecule**.

1. Use the sliders to set the number of **O₂ molecules** and **H₂ molecules** to two.

A. How many hydrogen molecules (H₂) are there?

Hydrogen atoms?

B. How many oxygen molecules (O₂) are there?

Oxygen atoms?

2. How many H₂O molecules do you think will form when these four molecules react?

3. Click **Play** (▶). How many H₂O molecules actually formed?



Limiting Reactant Gizmo Answer Key

Sebastian Brünink



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Table of Contents Limiting Reactant Gizmo Answer Key

1. Understanding the eBook Limiting Reactant Gizmo Answer Key
 - The Rise of Digital Reading Limiting Reactant Gizmo Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Limiting Reactant Gizmo Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Limiting Reactant Gizmo Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Limiting Reactant Gizmo Answer Key
 - Personalized Recommendations
 - Limiting Reactant Gizmo Answer Key User Reviews and Ratings
 - Limiting Reactant Gizmo Answer Key and Bestseller Lists
5. Accessing Limiting Reactant Gizmo Answer Key Free and Paid eBooks
 - Limiting Reactant Gizmo Answer Key Public Domain eBooks

- Limiting Reactant Gizmo Answer Key eBook Subscription Services
- Limiting Reactant Gizmo Answer Key Budget-Friendly Options
- 6. Navigating Limiting Reactant Gizmo Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Limiting Reactant Gizmo Answer Key Compatibility with Devices
 - Limiting Reactant Gizmo Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Limiting Reactant Gizmo Answer Key
 - Highlighting and Note-Taking Limiting Reactant Gizmo Answer Key
 - Interactive Elements Limiting Reactant Gizmo Answer Key
- 8. Staying Engaged with Limiting Reactant Gizmo Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Limiting Reactant Gizmo Answer Key
- 9. Balancing eBooks and Physical Books Limiting Reactant Gizmo Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Limiting Reactant Gizmo Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Limiting Reactant Gizmo Answer Key
 - Setting Reading Goals Limiting Reactant Gizmo Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Limiting Reactant Gizmo Answer Key
 - Fact-Checking eBook Content of Limiting Reactant Gizmo Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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