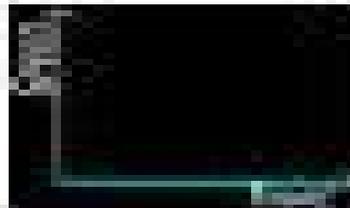
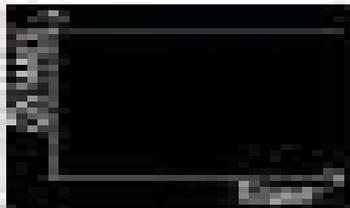


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 CH101

This assignment will require you to use your knowledge of the following topics to answer the questions. You should refer to your lecture notes and any relevant textbooks. You should also refer to the relevant chapters in the following books: Principles of Chemistry by G. L. Fisher and Principles of Chemistry by R. A. Scott.

Questions

- 1) Define the following terms: **enthalpy** **entropy** **free energy** **equilibrium constant**
- 2) Define the difference between standard enthalpy of formation and standard enthalpy of combustion.  
 Give the standard enthalpy of formation of **ethanol** and **ethane**.
- 3) Define the difference between standard enthalpy of formation and standard enthalpy of combustion.  
 Give the standard enthalpy of formation of **ethanol** and **ethane**.
- 4) Explain the difference between standard enthalpy of formation and standard enthalpy of combustion.  
 Give the standard enthalpy of formation of **ethanol** and **ethane**.
- 5) Define the difference between standard enthalpy of formation and standard enthalpy of combustion.  
 Give the standard enthalpy of formation of **ethanol** and **ethane**.
- 6) Define the difference between standard enthalpy of formation and standard enthalpy of combustion.  
 Give the standard enthalpy of formation of **ethanol** and **ethane**.
- 7) Define the difference between standard enthalpy of formation and standard enthalpy of combustion.  
 Give the standard enthalpy of formation of **ethanol** and **ethane**.
- 8) Define the difference between standard enthalpy of formation and standard enthalpy of combustion.  
 Give the standard enthalpy of formation of **ethanol** and **ethane**.



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