

# CHEMISTRY TEST STUDY GUIDE ANSWER KEY

Complete the following information about atoms.

Label the subatomic particles in the atom diagram below.



What charge does each subatomic particle have?

Proton: **positive**

Electron: **negative**

Neutron: **neutral (no charge)**

What subatomic particles are found in the nucleus?

**protons and neutrons**

Classify each of the following substances as an Element (E), Compound (C), or Mixture (M).

*If it's a mixture, is it heterogeneous (he) or homogeneous (ho)?*

- water **C**
- salt **C**
- air **M (ho)**
- carbon **E**
- sugar **C**
- saltwater **M (ho)**
- hydrogen **E**
- carbon dioxide **C**
- kool aid **M (ho)**

Use a periodic table to answer the following questions.

- Is sodium a metal or nonmetal? **metal**
- What is the atomic number of oxygen? **8**
- What is element symbol for iron? **Fe**
- What other elements are in the same family as Chlorine? **Fluorine, Bromine, Iodine, Astatine, Tennessine**
- Li, Be, B, C, N, O, F, and Ne are all in the same: **Period**

# Glencoe Science Chemistry Chapter 23 Study Guide Answer Key

**Jessica J Manson**



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