Q1

The common oxidation states of Ti are

(a) +2 and +3 (b) +3 and +4
(c) −3 and −4 (d) +2, +3 and +4

Q2

When phenol is treated with excess bromine water. It gives

(a) *m*-Bromophenol (b) *o*-and *p*-Bromophenols
(c) 2,4-Dibromophenol (d) 2,4, 6-Tribromophenol.

Q3

If 0.5 mol of $BaCl_2$ is mixed with 0.2 mole of Na_3PO_4, find the maximum amount of $Ba_3(PO_4)_2$ that can be formed.

(a) 1 mole (b) 0.5 mole (c) 0.1 mole (d) 0.01 mole

Q4

Which is chemically most active non-metal ?

(a) S (b) O (c) F (d) N

Q5

The following species will not exhibit disproportionation reaction

(a) ClO^- (b) ClO_2^- (c) ClO_3^- (d) ClO_4^-

Q6

Which of the following has been used in the manufacture of non-inflammable photographic films?

(a) Cellulose nitrate (b) Cellulose xanthate
(c) Cellulose perchlorate (d) Cellulose acetate

Q7

Point out the wrong statement:
For a first order reaction

(a) time for half-change ($t_{1/2}$) is independent of initial concentration
(b) change in the concentration unit does not change the rate constant (k)
(c) time for half-change × rate constant = 0.693
(d) the unit of k is $mol^{-1}\ min^{-1}$

Q8

Name of the structure of silicates in which three oxygen atoms of $[SiO_4]^{4-}$ are shared.

(a) Pyrosilicate
(b) Sheet silicate
(c) Linear chain silicate
(d) Three dimensional silicate

Q9

In the following sequence of reactions

$$C_2H_5Br \xrightarrow{AgCN} X \xrightarrow{\text{Reduction}} Y$$: Y is

(a) *n*-propyl amine (b) isopropylamine
(c) ethylamine (d) ethylmethyl amine

Q10

Which of the following monosaccharide is a pentose ?

(a) Galactose (b) Glucose
(c) Fructose (d) Arabinose

Q11

In Bohr's model, atomic radius of the first orbit is y, the radius of the 3rd orbit, is

(a) $y/3$ (b) y
(c) $3y$ (d) $9y$

Q12

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Shasha Hu

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