

General Chemistry II Jasperse  
Acid-Base Chemistry: Extra Practice Problems

General Types/Groups of problems:

Conceptual Questions: Acids, Bases, and Conjugates, Miscellaneous	p1	$K_b$ and $pK_b$ , Base Strength, and using $K_b$ or $pK_b$ to Calculate $[OH^-]$ , $pOH$ , $pH$ , and/or $[H^+]$	p7-10
Recognizing Strong versus Weak Acids; Recognizing Basic versus Nonbasic	p3	Recognizing Acid/Base Properties when Ionics are Dissolved in Water	p11
pH Calculations: Relationships between pH and pOH	p4	Answers	p12
$K_a$ , Sense + Calculations: Using $K_a$ or $pK_a$ to Calculate $[H^+]$ and/or pH; using pH to calculate $K_a$ or $pK_a$	p5		

**Conceptual Questions: Acids, Bases, and Conjugates, Miscellaneous**

- In the Brønsted-Lowry definition of acids and bases, an acid \_\_\_\_\_
  - is a proton donor.
  - is a proton acceptor.
  - forms stable hydrogen bonds.
  - breaks stable hydrogen bonds.
  - corrodes metals.
- In the Brønsted-Lowry definition of acids and bases, a base \_\_\_\_\_
  - is a proton donor.
  - is a proton acceptor.
  - forms stable hydrogen bonds.
  - breaks stable hydrogen bonds.
  - corrodes metals.
- In the following reaction in aqueous solution, the acid reactant is \_\_\_\_\_ and its conjugate base product is \_\_\_\_\_
 
$$CH_3COOH + NH_3 \rightleftharpoons CH_3COO^- + NH_4^+$$
  - $CH_3COOH$ ;  $CH_3COO^-$
  - $CH_3COOH$ ;  $NH_4^+$
  - $NH_3$ ;  $CH_3COO^-$
  - $NH_3$ ;  $NH_4^+$
  - $CH_3COOH$ ;  $H_3O^+$
- In the following reaction in aqueous solution, the acid reactant is \_\_\_\_\_, and its conjugate base product is \_\_\_\_\_
 
$$CH_3NH_2 + HSO_4^- \rightleftharpoons CH_3NH_3^+ + SO_4^{2-}$$
  - $CH_3NH_2$ ;  $CH_3NH_3^+$
  - $CH_3NH_2$ ;  $SO_4^{2-}$
  - $HSO_4^-$ ;  $CH_3NH_3^+$
  - $HSO_4^-$ ;  $SO_4^{2-}$
  - $HSO_4^-$ ;  $H_3O^+$
- Which of the following is the conjugate acid of the hydrogen phosphate ion,  $HPO_4^{2-}$ ?
  - $H_2PO_4^-$
  - $H_3PO_4$
  - $HPO_4^{2-}$
  - $PO_4^{3-}$
  - $H_3O^+$

# General Chemistry Acid Base Practice Problems

**Xiaolong Qi**



**General Chemistry Acid Base Practice Problems:**

## Whispering the Secrets of Language: An Mental Journey through **General Chemistry Acid Base Practice Problems**

In a digitally-driven world wherever screens reign great and immediate connection drowns out the subtleties of language, the profound secrets and mental subtleties hidden within words often move unheard. However, situated within the pages of **General Chemistry Acid Base Practice Problems** a captivating literary treasure sporting with raw thoughts, lies a fantastic quest waiting to be undertaken. Composed by an experienced wordsmith, that enchanting opus invites viewers on an introspective trip, delicately unraveling the veiled truths and profound impact resonating within the material of every word. Within the emotional depths of this emotional review, we can embark upon a heartfelt exploration of the book is key subjects, dissect their interesting writing design, and succumb to the effective resonance it evokes serious within the recesses of readers hearts.

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